

## Review

# Exploration of new superconductors and functional materials, and fabrication of superconducting tapes and wires of iron pnictides

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## Abstract

This review shows the highlights of a 4-year-long research project supported by the Japanese Government to explore new superconducting materials and relevant functional materials. The project found several tens of new superconductors by examining ~1000 materials, each of which was chosen by Japanese experts with a background in solid state chemistry. This review summarizes the major achievements of the project in newly found superconducting materials, and the fabrication wires and tapes of iron-based superconductors; it incorporates a list of ~700 unsuccessful materials examined for superconductivity in the project. In addition, described are new functional materials and functionalities discovered during the project.

Keywords: superconductivity, iron pnictide, new superconductors, superconducting wire, superconducting tape, functional material, powder in tube



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## 1. Introduction

Since the discovery of superconductivity by Heike Kamerling Onnes in 1911 [1], it has been the dream of researchers to realize a room temperature superconductor. Although the fundamental theoretical framework for superconductivity was

**Table 2.** Exploration of new superconductors in this project. (a) Materials which exhibited superconductivity. (b) Materials which exhibited no superconductivity.

(a) Materials which exhibited superconductivity

noted in this review, (HP); synthesized using high pressure technique

IBSC			
	Material	$T_c$ (K)	Comment
1	SmFeAsO <sub>1-x</sub> H <sub>x</sub> [9]	56	large solubility limit of dopant H <sup>+</sup> (HP)
2	NdFeAsO <sub>1-x</sub> H <sub>x</sub> [9]	54	large solubility limit of dopant H <sup>+</sup> (HP)
3	CeFeAsO <sub>1-x</sub> H <sub>x</sub> [10]	48	large solubility limit of dopant H <sup>+</sup> (HP)
4	LaFeAsO <sub>1-x</sub> H <sub>x</sub> [11]	36	two $T_c$ domes (HP)
5	SmFeAs <sub>1-x</sub> P <sub>x</sub> O <sub>1-x</sub> H <sub>x</sub> [12]	<56	$T_c$ decreased by replacing with P (HP)
6	LaFe <sub>1-x</sub> Zn <sub>x</sub> AsO <sub>1-x</sub> H <sub>x</sub> [13]	<36	$T_c$ decreased by replacing with Zn (HP)
7	LaFeAsO <sub>0.85</sub> H <sub>x</sub> ( $x=0-0.85$ ) [14]	35	indirect electron doping by $V_C$ and $V'$ (HP)
8	LaFeAsO <sub>1-x</sub> [15]	22	indirect electron doping by $V_C$ (HP)
9	CaFe <sub>1-x</sub> Co <sub>x</sub> AsH [16]	23	direct electron doping into CaFeAsH (HP)
10	Ca <sub>1-x</sub> La <sub>x</sub> FeAsH [17]	47	indirect electron doping into CaFeAsH (HP)
11	Ca <sub>1-x</sub> Sm <sub>x</sub> FeAsH [17]	46	indirect electron doping into CaFeAsH (HP)
12	CaFeAsF <sub>1-x</sub> [18]	27	indirect electron doping (HP)
13	Sr <sub>1-x</sub> La <sub>x</sub> Fe <sub>2</sub> As <sub>2</sub> [19]	23	indirect electron doping (HP)
14	(Ba,La)Fe <sub>2</sub> As <sub>2</sub> [20]	22.4	indirect electron doping (thin film)
15	(Ba,Ce)Fe <sub>2</sub> As <sub>2</sub> [21]	13.5	indirect electron doping (thin film)
16	(Ba,Pr)Fe <sub>2</sub> As <sub>2</sub> [21]	6.1	indirect electron doping (thin film)
17	(Ba,Nd)Fe <sub>2</sub> As <sub>2</sub> [21]	5.8	indirect electron doping (thin film)
18	(Sr,La)Fe <sub>2</sub> As <sub>2</sub> [22]	20	indirect electron doping (thin film)
19	(Ca,La)Fe <sub>2</sub> (As,P) <sub>2</sub> [23]	45	bulk SC achieved by co-doping of La and P
20	Ba(Fe,Pt) <sub>2</sub> As <sub>2</sub> [24]	24	dome-like relation between $T_c$ and dopant Pt
21	(Ca,La)FeAs <sub>2</sub> [25]	34	new type IBSC: 112
22	(Ca,La)Fe(As,Sb) <sub>2</sub> [26]	43	increase in $T_c$ by co-doping
23	(Ca,RE)Fe(As,Sb) <sub>2</sub> [27]	47	highest $T_c$ in 112 type ( $RE=La, Ce, Pr, Nd$ )
24	Ca <sub>1-x</sub> (Ir <sub>x</sub> As <sub>1-x</sub> )(Fe <sub>2</sub> As <sub>2</sub> ) <sub>2</sub> [28]	16	analogous to Ca <sub>1-x</sub> (Pt <sub>x</sub> As <sub>1-x</sub> )(Fe <sub>2</sub> Pt <sub>1-x</sub> As <sub>2</sub> ) <sub>2</sub>
25	Na <sub>0.85</sub> Fe <sub>1.85</sub> Se <sub>2</sub> [29]	37	intercalating Na into FeSe
26	(Na,NH <sub>3</sub> )Fe <sub>2</sub> Se <sub>2</sub> [29]	42	co-intercalating Na and NH <sub>3</sub> into FeSe
27	LaFeAs(O,C) <sub>2</sub>	27	electron doping by C <sub>2</sub> (HP)
Analogous structure to IBSC			
	Material	$T_c$ (K)	Comment
28	SrAl <sub>2</sub> Si <sub>2</sub>	4.6	IBSC 122 type structure (HP)
29	NbSiAs [30]	8.2	square net of Si
30	BaTi <sub>2</sub> Sb <sub>2</sub> O [31]	3	square net of Ti (measured under HP)
31	BaTi <sub>2</sub> (Sb <sub>1-x</sub> Bi <sub>x</sub> ) <sub>2</sub> O [32]	1.2	square net of Ti, SC by doping of isovalent ion
32	BaTi <sub>2</sub> (Sb <sub>1-x</sub> Sn <sub>x</sub> ) <sub>2</sub> O [33]	2.5	square net of Ti, SC by doping of aliovalent ion
33	La(Co <sub>1-x</sub> Fe <sub>x</sub> ) <sub>2</sub> B <sub>2</sub> [34]	4	square net of Co
34	(La <sub>1-x</sub> Y <sub>x</sub> )Co <sub>2</sub> B <sub>2</sub> [34]	4	square net of Co
35	BaNi <sub>2</sub> (As,P) <sub>2</sub> [35]	3.3	analogous to 122 type of IBSC
36	Ba(Ni,Cu) <sub>2</sub> As <sub>2</sub>	3.3	analogous to 122 type of IBSC
37	L <sub>n</sub> NiAsO <sub>1-x</sub> H <sub>x</sub> [36]	3.7	H doping to LnNiAsO ( $L_n=La-Nd$ ) (HP)
38	NdNi <sub>0.64</sub> Bi <sub>2</sub>	4	square net of Ni and Bi
39	YNi <sub>2</sub> Bi <sub>2</sub> [37]	4	square net of Ni and Bi
40	LaNi <sub>2</sub> Bi <sub>2</sub> [37]	4	square net of Ni and Bi
41	CeNi <sub>2</sub> Bi <sub>2</sub> [37]	4	square net of Ni and Bi
42	LaNiBN [38]	4.1	square net of Ni (HP)
43	CaNiBN [38]	2.2	square net of Ni (HP)
44	LaPtBN [38]	6.7	square net of Pt (HP)
45	La <sub>3</sub> Ni <sub>2</sub> B <sub>2</sub> N <sub>3</sub> [38]	15	square net of Ni (HP)
46	LaPd <sub>2</sub> As <sub>2</sub> [39]	1	collapsed 122 structure
47	LaPd <sub>2</sub> Sb <sub>2</sub> [40]	1.4	CaBe <sub>2</sub> Ge <sub>2</sub> structure
48	SrPt <sub>2</sub> Sb <sub>2</sub> [41]	2.1	CaBe <sub>2</sub> Ge <sub>2</sub> structure
49	BaPt <sub>2</sub> Sb <sub>2</sub> [42]	1.9	CaBe <sub>2</sub> Ge <sub>2</sub> structure
50	SrPd <sub>2</sub> Bi <sub>2</sub>	2.2	CaBe <sub>2</sub> Ge <sub>2</sub> structure
51	CaPd <sub>2</sub> Bi <sub>2</sub>	2.6	CaBe <sub>2</sub> Ge <sub>2</sub> structure
52	SrPt <sub>2</sub> Bi <sub>2</sub>	2.6	CaBe <sub>2</sub> Ge <sub>2</sub> structure
53	La <sub>3</sub> Sb [43]	5	square net of La
Other layered structure			
	Material	$T_c$ (K)	Comment
54	La <sub>3</sub> Ru <sub>2</sub> B <sub>6</sub> [44]	3.2	homologous structure
55	CrNbN	11	Cr layer (multiphase)

Table 2. (Continued.)

56	Ca <sub>2</sub> Al <sub>2</sub> Si <sub>4</sub> [45]	6.4	high pres. phase of Ca-Al-Si (HP)
57	Mg <sub>2</sub> AlSi <sub>3</sub> [46]	5.2	new layered structure (HP)
58	Sr(Si <sub>1-x</sub> Ni <sub>x</sub> ) <sub>2</sub>	3	Si layer
59	Ba(Si <sub>1-x</sub> Cu <sub>x</sub> ) <sub>2</sub>	3	Si layer
60	LaSi <sub>2</sub> H <sub>0.03</sub>	3	H doping to ThSi <sub>2</sub> structure ( <i>T<sub>c</sub></i> increased)
61	BaGe <sub>2</sub> H <sub>0.17</sub>	4	H doping to ThSi <sub>2</sub> structure ( <i>T<sub>c</sub></i> decreased)
62	LaGe <sub>1.7</sub>	2	ThSi <sub>2</sub> structure with cage formed by sp <sup>3</sup> orbital
63	Zr <sub>2</sub> Ru <sub>2</sub> Si <sub>4</sub> [47]	5.7	analogous structure to ZrRuP superconductor
64	MgPtSi [48]	2.5	honeycomb lattice
65	SrAuSi <sub>3</sub> [49]	1.54	noncentrosymmetric structure
66	Li <sub>2</sub> IrSi <sub>3</sub> [50]	3.8	noncentrosymmetric structure
67	LaIrPn [51]	5.3	noncentrosymmetric structure ( <i>P<sub>n</sub></i> = P and Δs)
68	LaRhP [51]	2.5	noncentrosymmetric structure
69	Bi <sub>2</sub> O <sub>4</sub> S <sub>3</sub> [52]	4.5	double BiS plane
70	LnO <sub>1-x</sub> F <sub>x</sub> BiS <sub>3</sub> (Ln=La,Ce,Nd) [53]	3 (10)	triple BiS plane. (measured under HP)
71	KMo <sub>3</sub> Se <sub>6</sub>	9	new Chevrel phase compound
72	Nb <sub>2</sub> IrO	10.5	O doped Mn <sub>2</sub> Si <sub>3</sub> structure
73	Ba <sub>2</sub> (Bi <sub>1-x</sub> Sb <sub>x</sub> ) <sub>3</sub> [54]	4.3	square/honeycomb lattice
74	Y <sub>2</sub> CrC <sub>3</sub>	1.5	Cr/C layer
75	SrNiSn <sub>3</sub>	5	noncentrosymmetric structure
76	β-ZrNF	15	ion exchange of β-ZrNCl by F
77	Zr <sub>2</sub> (N <sub>2</sub> O)F <sub>6</sub>	2-5	low SVF
<b>Intercalation to layered structure</b>			
	Material	<i>T<sub>c</sub></i> (K)	Comment
78	TiNCl← amines [55]	17	intercalation of long alkylene diamine → high <i>T<sub>c</sub></i>
79	TiNCl← alkali metal (Li-Rb) [56]	18.0	highest <i>T<sub>c</sub></i> by Na intercalation
80	TiNCl← alkali metal + solvent [56]	10.5	highest <i>T<sub>c</sub></i> by Na <sup>+</sup> THF co-intercalation
81	TiNB← alkali metal [57]	17.2	highest <i>T<sub>c</sub></i> by K intercalation
82	ZrNCl← K + THF, PC [58]	16	highest <i>T<sub>c</sub></i> by K+THF co-intercalation
83	α-ZrNCl← alkali metal	10	superconductivity of α-type structure
84	HfNCl← AE(Ca-Ba), solvent [59]	26	highest <i>T<sub>c</sub></i> by Ca+THF co-intercalation
85	MNCl← Yb, Eu (M: Zr, Hf) [60]	24.3	highest <i>T<sub>c</sub></i> by Eu <sup>+</sup> NH <sub>3</sub> co-intercalation into HfNCl
86	Hf <sub>2</sub> N <sub>2</sub> O← Li	4.6	La <sub>2</sub> O <sub>3</sub> type structure
<b>Chalcogenides</b>			
	Material	<i>T<sub>c</sub></i> (K)	Comment
87	1T-TaS <sub>2</sub>	3	suppressing CDW (HP)
88	AuTe <sub>2</sub> [61]	2.3	SC by dissociation of Te <sub>2</sub> dimer (CdI <sub>2</sub> structure)
89	(Au,Pt)Te <sub>2</sub> [62]	4	SC by dissociation of Te <sub>2</sub> dimer (CdI <sub>2</sub> structure)
90	(Au,Pd)Te <sub>2</sub>	3	SC by dissociation of Te <sub>2</sub> dimer (CdI <sub>2</sub> structure)
91	(Ir,Pt)Te <sub>2</sub> [63]	3.1	SC by dissolving of orbital ordering (CdI <sub>2</sub> structure)
92	Cu <sub>2</sub> IrTe <sub>2</sub>	3	SC by dissolving of orbital ordering (CdI <sub>2</sub> structure)
93	(Ir,Rh)Te <sub>2</sub> [64]	2.6	SC by dissolving of orbital ordering (CdI <sub>2</sub> structure)
94	Ir <sub>2</sub> Se <sub>3</sub> [65]	8	pyrite structure (HP)
95	Ir <sub>1-x</sub> Rh <sub>x</sub> Se <sub>2</sub> [66]	10	pyrite structure (Rh doping) (HP)
96	Ir <sub>0.95</sub> Rh <sub>0.05</sub> Te <sub>2</sub> [67]	4.6	pyrite structure (small effect of Rh doping) (HP)
<b>Others</b>			
	Material	<i>T<sub>c</sub></i> (K)	Comment
97	NbBe <sub>2</sub>	2.6	Laves phase
98	MoC <sub>0.68</sub> [68]	12	formed MoC <sub>0.681</sub> under 6 GPa
99	MoC <sub>0.75</sub> [69]	13	formed MoC <sub>0.75</sub> under 17 GPa
100	YFe <sub>2</sub> SiC	3.5	having YFe <sub>2</sub> Si framework
101	SeC <sub>2</sub> O <sub>2</sub>	8	having C-C bonding (low SVF)
102	Ti <sub>4-x</sub> N <sub>3</sub>	<5	decomposition of TiNCl in NH <sub>3</sub>
103	Zr <sub>4-x</sub> N <sub>3</sub>	<10	decomposition of ZrNCl in NH <sub>3</sub>
104	Hf <sub>4-x</sub> N <sub>3</sub>	<6	decomposition of HfNCl in NH <sub>3</sub>
105	Mg(Mg <sub>1-x</sub> Al <sub>x</sub> )Si [70]	6.2	anticotunnite or TiN(Si) structure (HP)
106	Ca(Al, Si) <sub>2</sub> [71]	2.6	Laves phase (HP)
107	Nb <sub>2</sub> MSi (M=Ni, Co, Fe) [72]	7.7	2 dimensional network of Nb
108	Ba <sub>2</sub> Ir <sub>2</sub> Ge <sub>24</sub> [73]	6.1	aving cage structure
109	Ba <sub>2</sub> Ir <sub>2</sub> Ge <sub>24</sub> [73]	3.2	having cage structure
110	Ca <sub>2</sub> InN [74]	0.6	doping of In <sup>3+</sup> into Ca <sub>2</sub> N
111	SnAs <sub>8</sub>	2	hole in As4p orbital
112	CuZr <sub>2</sub>	1	
113	HfZr <sub>2</sub>	1	

Table 2. (Continued.)

(b) Materials which exhibited no superconductivity

: possible to obtain the target phase,  : impossible to obtain the target phase

IBSC (Square net of Fe)				
MgFeAsH	SrFeAsH	BaFeAsH	EuFeAsH	MgFePH
CaFePH	CaFeAsF <sub>1-x</sub>	CaFeAsF <sub>1-x</sub> O <sub>x</sub>	CaFeAsF <sub>1-x</sub> H <sub>x</sub> [9]	Ca <sub>1-x</sub> La <sub>x</sub> FeAsF
Sr <sub>1-x</sub> La <sub>x</sub> FeAsF	Ca <sub>1-x</sub> Na <sub>x</sub> FeAsH	LaFeAsO:N	CaFeAsH <sub>1-x</sub> O <sub>x</sub>	LaFeOsB
CaFeOSe	SrFeOSe	KFeSeF	LaFeSiF	Ba(Fe,Re) <sub>2</sub> As <sub>2</sub>
Ca(Fe,Au) <sub>2</sub> As <sub>2</sub> (thin film)	Ca(Fe,Mg) <sub>2</sub> As <sub>2</sub>	Ca(Fe,Al) <sub>2</sub> As <sub>2</sub>	Ca(Fe,Pt) <sub>2</sub> As <sub>2</sub> [75]	Ca(Fe,Ru) <sub>2</sub> As <sub>2</sub>
(Ca,La)Fe <sub>2</sub> As <sub>2</sub>	YFe <sub>2-x</sub> Co <sub>x</sub> Ge <sub>2</sub>	BaFe <sub>2-x</sub> Zn <sub>x</sub> As <sub>2</sub>	BaFe <sub>2</sub> Sb <sub>2</sub>	BaFe <sub>2</sub> Se <sub>2</sub>
(Fe,Mn) <sub>2</sub> AlB <sub>2</sub>	(Fe,Co) <sub>2</sub> AlB <sub>2</sub>	(Fe,Ni) <sub>2</sub> AlB <sub>2</sub>	(Fe,Cr) <sub>2</sub> AlB <sub>2</sub>	ZrFe <sub>2</sub> Si <sub>2</sub>
ZrFe <sub>2</sub> B <sub>2</sub>	LuFe <sub>2-x</sub> Si <sub>x</sub>	YFe <sub>2-x</sub> Si <sub>x</sub>	TlFe <sub>2-x</sub> Co <sub>x</sub> Se <sub>2</sub>	TlFe <sub>2-x</sub> Ni <sub>x</sub> Se <sub>2</sub>
Tl <sub>1-x</sub> Fe <sub>x</sub> Se <sub>2</sub>	MgFe <sub>2</sub> As <sub>2</sub>	CdFe <sub>2</sub> As <sub>2</sub>	YFe <sub>2</sub> B <sub>2</sub>	Fe <sub>2-x</sub> Cu <sub>x</sub> As
(Fe,Pt)Se	Fe(Te,As)	Fe(Se,As)	FeP <sub>0.5</sub> X <sub>0.5</sub> (X = Cl, Br)	FeSe <sub>1-x</sub> X <sub>x</sub> (X = P, Cl)
FeSb <sub>1-x</sub>	La-Ca <sub>10</sub> (Ir <sub>4</sub> As <sub>8</sub> )(Fe <sub>2</sub> As <sub>2</sub> ) <sub>3</sub>	Ge-Ca <sub>10</sub> (Ir <sub>4</sub> As <sub>8</sub> )(Fe <sub>2</sub> As <sub>2</sub> ) <sub>3</sub>	P-Ca <sub>10</sub> (Ir <sub>4</sub> As <sub>8</sub> )(Fe <sub>2</sub> As <sub>2</sub> ) <sub>3</sub>	Re-Ca <sub>10</sub> (Ir <sub>4</sub> As <sub>8</sub> )(Fe <sub>2</sub> As <sub>2</sub> ) <sub>3</sub>
K-Ca <sub>10</sub> (Ir <sub>4</sub> As <sub>8</sub> )(Fe <sub>2</sub> As <sub>2</sub> ) <sub>3</sub>	Sr-Ca <sub>10</sub> (Ir <sub>4</sub> As <sub>8</sub> )(Fe <sub>2</sub> As <sub>2</sub> ) <sub>3</sub>	Ba-Ca <sub>10</sub> (Ir <sub>4</sub> As <sub>8</sub> )(Fe <sub>2</sub> As <sub>2</sub> ) <sub>3</sub>	Ca <sub>10</sub> (M <sub>4</sub> As <sub>8</sub> )(Fe <sub>2</sub> As <sub>2</sub> ) <sub>3</sub> (M = Pd, Rh)	Ba <sub>10</sub> (Pt <sub>4</sub> As <sub>8</sub> )(Fe <sub>2</sub> As <sub>2</sub> ) <sub>3</sub>
Sr <sub>10</sub> (Pt <sub>4</sub> As <sub>8</sub> )(Fe <sub>2</sub> As <sub>2</sub> ) <sub>3</sub>	Ca <sub>10</sub> Pt <sub>3.8</sub> Fe <sub>0.2</sub> Sb <sub>18</sub>	Ba <sub>10</sub> IrO <sub>3</sub> FeAs	Sr <sub>2</sub> InO <sub>3</sub> FeAs	Sr <sub>2</sub> FeO <sub>3</sub> FeAs
Ba <sub>3</sub> FeO <sub>3</sub> FeAs	Sr <sub>2</sub> V <sub>2</sub> O <sub>3</sub> Fe <sub>2</sub> As <sub>2</sub>	Bi <sub>2</sub> SrFe <sub>2</sub> O <sub>3</sub> Se <sub>2</sub>	La <sub>3</sub> O <sub>3</sub> Fe <sub>2</sub> As <sub>4</sub>	Ca <sub>3</sub> F <sub>2</sub> Fe <sub>4</sub> As <sub>2</sub>
La <sub>2</sub> O <sub>3</sub> Fe <sub>2</sub> (Se <sub>1-x</sub> F <sub>x</sub> ) <sub>2</sub>	La <sub>2</sub> O <sub>3</sub> Fe <sub>2</sub> (Se <sub>1-x</sub> F <sub>x</sub> ) <sub>2</sub>	La <sub>2</sub> O <sub>3</sub> Fe <sub>2</sub> (Se <sub>1-x</sub> As <sub>x</sub> ) <sub>2</sub>	La <sub>2</sub> O <sub>3</sub> Fe <sub>2</sub> (Se <sub>1-x</sub> Ge <sub>x</sub> ) <sub>2</sub>	La <sub>2</sub> O <sub>3</sub> (Fe <sub>1-x</sub> Co <sub>x</sub> ) <sub>2</sub> Se
La <sub>2</sub> O <sub>3</sub> (Fe <sub>1-x</sub> Mn <sub>x</sub> ) <sub>2</sub> Se	NdFe <sub>2</sub> Sb <sub>2</sub>	Fe <sub>1-x</sub> Cu <sub>x</sub> As <sub>0.5</sub>	EuFeAsO <sub>1-x</sub>	Fe <sub>1-x</sub> Si <sub>x</sub>
(Fe <sub>1-x</sub> Mn <sub>x</sub> ) <sub>2</sub> Si <sub>2</sub>	(Fe <sub>1-x</sub> Co <sub>x</sub> ) <sub>2</sub> Si <sub>2</sub>	(Fe <sub>1-x</sub> Mo <sub>x</sub> ) <sub>2</sub> Si <sub>2</sub>	AgFeAs	La(Fe <sub>1-x</sub> Co <sub>x</sub> ) <sub>2</sub> Si <sub>2</sub>
Fe <sub>0.85</sub> (Si <sub>1-x</sub> Ge <sub>x</sub> ) <sub>2</sub>	RFeC <sub>2</sub>	Se(Fe,Co)C <sub>2</sub>	Sc(Fe,Ni)C <sub>7</sub>	La(Fe,Co)Si <sub>2</sub>
La(Fe,Mn)Si <sub>2</sub>	LaFe <sub>0.5</sub> Sb <sub>2</sub>	CaFe <sub>4</sub> As <sub>3</sub>	CaFe <sub>3</sub> AgAs <sub>3</sub>	CaFe <sub>3</sub> CuAs <sub>3</sub>
CaFe <sub>3</sub> LiAs <sub>3</sub>	La <sub>3</sub> O <sub>3</sub> Fe <sub>4</sub> As <sub>3</sub>	Fe(Se,As) <sub>2</sub>	(Fe,Mn)As <sub>2</sub>	(Fe,Ru)As <sub>2</sub>
(Fe,Rh)As <sub>2</sub>	(Fe,Co)As <sub>2</sub>	(Fe,Ni)As <sub>2</sub>	(Fe,Pd)As <sub>2</sub>	La <sub>1-x</sub> RE <sub>x</sub> Cu <sub>1-x</sub> O <sub>0.8</sub> Fe <sub>2</sub> As <sub>2</sub>
Sr <sub>2</sub> Fe <sub>2</sub> Cu <sub>2</sub> Se <sub>2</sub> O <sub>3</sub>	Sr <sub>2-x</sub> K <sub>0.2</sub> Fe <sub>2</sub> Cu <sub>2</sub> Se <sub>2</sub> O <sub>3</sub>	La <sub>3</sub> O <sub>3</sub> Fe <sub>4</sub> As <sub>4</sub>	Ca <sub>3</sub> Fe <sub>4</sub> As <sub>4</sub> F <sub>3</sub>	Ca <sub>3</sub> FeO <sub>6</sub>
Ba <sub>1-x</sub> K <sub>x</sub> Fe <sub>2</sub> X (X = S, Se)	NaFe <sub>2</sub> O <sub>4</sub>	BaFe <sub>2-x</sub> Co <sub>x</sub> Se <sub>3</sub>	Tl <sub>1-x</sub> Ba <sub>x</sub> FeSe <sub>2</sub>	La <sub>2</sub> Fe <sub>2</sub> Se <sub>2</sub> O <sub>1-x</sub> H <sub>x</sub>
La <sub>3</sub> Fe <sub>2</sub> S <sub>2</sub> O <sub>1-x</sub> H <sub>x</sub>	CaBaFe <sub>2</sub> Pn <sub>2</sub> O (Pn = As, Sb, Bi)	SrFe <sub>2</sub> Pn <sub>2</sub> O (Pn = As, Sb, Bi)	BaFe <sub>2</sub> Pn <sub>2</sub> O (Pn = As, Sb, Bi)	BaFe <sub>2</sub> Se <sub>2</sub> O
t-FeS				
Analogous structure to IBSC material				
Square net of transition metal elements		BaSc <sub>2</sub> Pn <sub>2</sub> O (Pn = As, Sb, Bi)	SrTi <sub>2</sub> As <sub>2</sub> O	BaTi <sub>2</sub> (As,Sb) <sub>2</sub> O [31]
EuTi <sub>2</sub> As <sub>2</sub> O	BaTi <sub>2</sub> Sb <sub>2</sub> O	(SrF) <sub>2</sub> Ti <sub>2</sub> Bi <sub>2</sub> O [76]	(SrF) <sub>2</sub> Ti <sub>2</sub> (Sb <sub>1-x</sub> Bi <sub>x</sub> ) <sub>2</sub> O	Zr <sub>2</sub> Ti <sub>2</sub> As <sub>2</sub> H
Zr <sub>2</sub> Ti <sub>2</sub> Pn <sub>2</sub> H (Pn = Sb, Bi)	ZrTiPn (Pn = Sb, Bi)	ZrTiAs	BaV <sub>2</sub> Ge <sub>2</sub> O	BaV <sub>2</sub> Sn <sub>2</sub> O
BaV <sub>2</sub> Pn <sub>2</sub> O (Pn = As, Sb, Bi)	LnCrAsO (Ln = La-Nd, Sm) [77]	SrCrAsF	EuCrAsF	La <sub>1-x</sub> Ca <sub>x</sub> MnAsO <sub>1-x</sub> F <sub>x</sub>
Ce <sub>2-x</sub> Ca <sub>x</sub> MnAsO <sub>1-x</sub> F <sub>x</sub>	Sm <sub>1-x</sub> Ca <sub>x</sub> MnAsO <sub>1-x</sub> F <sub>x</sub>	SrMnAsF	BaMnAsF	EuMnAsF
LaMnAsO <sub>1-x</sub> H <sub>x</sub>	CeMnAsO <sub>1-x</sub> H <sub>x</sub>	DyMnAsO <sub>1-x</sub> H <sub>x</sub>	LaZn <sub>1-x</sub> Mn <sub>x</sub> AsO <sub>0.75</sub>	CaMn <sub>2</sub> As <sub>2</sub>
CaMn <sub>2</sub> Sb <sub>2</sub>	CaMnBi <sub>2</sub>	(Ca,La)Mn <sub>2</sub> Sb <sub>2</sub>	BaMn <sub>2</sub> As <sub>2</sub>	(Ba,K)Mn <sub>2</sub> As <sub>2</sub>
BaMnRuAs <sub>2</sub>	BaMnFeAs <sub>2</sub>	BaMnCoAs <sub>2</sub>	BaMnRhAs <sub>2</sub>	BaMnIrAs <sub>2</sub>
BaMnBi <sub>2</sub>	LaMn <sub>2</sub> Si <sub>2</sub>	Sr(Mn,Fe) <sub>2</sub> P <sub>2</sub>	Sr(Mn,Cr) <sub>2</sub> P <sub>2</sub>	Sr(Mn,Al) <sub>2</sub> P <sub>2</sub>
Sr(Mn,Zn) <sub>2</sub> P <sub>2</sub>	(Ca,Eu)Mn <sub>2</sub> P <sub>2</sub>	La <sub>2</sub> O <sub>3</sub> Mn <sub>2</sub> Te <sub>2</sub> O	CaMn <sub>2</sub> Pn <sub>2</sub> O (Pn = As, Sb, Bi)	SrMn <sub>2</sub> Pn <sub>2</sub> O (Pn = As, Sb, Bi)
BaMn <sub>2</sub> Pn <sub>2</sub> O (Pn = As, Sb, Bi)	BaMn <sub>2</sub> Se <sub>2</sub> O	Ca <sub>10</sub> (Pt <sub>4</sub> As <sub>8</sub> )(Mn <sub>2-x</sub> Pt <sub>x</sub> As <sub>2</sub> ) <sub>2</sub>	LaCoAsO <sub>1-x</sub> H <sub>x</sub>	(Ca,La)CoAs <sub>2</sub>
(Sr <sub>1-x</sub> Ca <sub>x</sub> )Co <sub>2</sub> Ge <sub>2</sub>	BaCo <sub>2</sub> Ge <sub>2</sub>	YCo <sub>2</sub> B <sub>2</sub>	YCo <sub>2</sub> Ge <sub>2</sub>	LaCo <sub>0.8</sub> Sb <sub>2</sub>
LaCo <sub>2</sub> Si <sub>2</sub>	La(Co <sub>1-x</sub> Ni <sub>x</sub> ) <sub>2</sub> Si <sub>2</sub>	LaCo <sub>2</sub> (Bi <sub>1-x</sub> Si <sub>x</sub> ) <sub>2</sub> [37]	LaCo <sub>2</sub> Ge <sub>2</sub>	NdCo <sub>2</sub> Sb <sub>2</sub>
GdCo <sub>2</sub> B <sub>2</sub>	LaCo <sub>2</sub> B <sub>2</sub> C <sub>2</sub>	ZrCo <sub>2</sub> Si <sub>2</sub>	HfCo <sub>2</sub> Si <sub>2</sub>	HfCo <sub>2</sub> Ge <sub>2</sub>
CaCo <sub>2</sub> Pn <sub>2</sub> O (Pn = As, Sb, Bi)	SrCo <sub>2</sub> Pn <sub>2</sub> O (Pn = As, Sb, Bi)	BaCo <sub>2</sub> Pn <sub>2</sub> O (Pn = As, Sb, Bi)	Ca <sub>10</sub> (Pt <sub>4</sub> As <sub>8</sub> )(Co <sub>2-x</sub> Pt <sub>x</sub> As <sub>2</sub> ) <sub>2</sub>	(Ca,La)NiAs <sub>2</sub>
CaNiAsH	Ba <sub>1-x</sub> La <sub>x</sub> Ni <sub>2</sub> As <sub>2</sub>	MgNiGe [78]	CaNi <sub>1-x</sub> Mn <sub>x</sub> Ge [79]	CaNi <sub>1-x</sub> Mn <sub>x</sub> GeH [80]
Y <sub>2</sub> NiSi <sub>2</sub>	LaNiAs	La <sub>2</sub> NiSi <sub>2</sub>	CeNiBiO <sub>1-x</sub>	ZrNi <sub>0.75</sub> P
ZrNi <sub>0.75</sub> As	KNi <sub>2</sub> Se <sub>2</sub>	ZrNi <sub>2</sub> Si <sub>2</sub>	RhNiP	GdNi <sub>2</sub> Bi <sub>2</sub>
Ni <sub>3</sub> (Te <sub>1-x</sub> Se <sub>x</sub> ) <sub>2</sub>	CaNi <sub>2</sub> Pn <sub>2</sub> O (Pn = As, Sb, Bi)	SrNi <sub>2</sub> Pn <sub>2</sub> O (Pn = As, Sb, Bi)	BaNi <sub>2</sub> Pn <sub>2</sub> O (Pn = As, Sb, Bi)	Ca <sub>10</sub> (Pt <sub>4</sub> As <sub>8</sub> )(Ni <sub>2-x</sub> Pt <sub>x</sub> As <sub>2</sub> ) <sub>3</sub>
Cu <sub>2</sub> As	Cu <sub>2</sub> Sb	MgCuAs	Ca <sub>2</sub> Cu <sub>2</sub> P <sub>3</sub>	BaCu <sub>1.5</sub> As <sub>2</sub>
BaCu <sub>2</sub> As <sub>2</sub>	Ba <sub>2</sub> Cu <sub>2</sub> P <sub>4</sub>	BaCu <sub>0</sub> As <sub>2</sub>	YCuSb <sub>2</sub>	YCuAs <sub>2</sub>
CeCuPO	GdCu <sub>1-2x</sub> P <sub>1-7x</sub>	ZrCu <sub>2</sub> P <sub>2</sub>	ZrCu <sub>2</sub> As <sub>2</sub>	ZrCuSiAs
HfCu <sub>2</sub> P <sub>2</sub>	Hf <sub>0.5</sub> Cu <sub>1.5</sub> Mn <sub>0.5</sub> P <sub>1</sub>	HfCuSi <sub>2</sub>	HfCuGe <sub>2</sub>	BiOCuS
Ca <sub>2</sub> Cu <sub>2</sub> Pn <sub>2</sub> O (Pn = As, Sb, Bi)	Sr <sub>2</sub> Cu <sub>2</sub> Pn <sub>2</sub> O (Pn = As, Sb, Bi)	Ba <sub>2</sub> Cu <sub>2</sub> Pn <sub>2</sub> O (Pn = As, Sb, Bi)	La <sub>3</sub> (O,F)Cu <sub>4</sub> As <sub>4</sub>	Bi <sub>2</sub> LnO <sub>4</sub> Cu <sub>2-x</sub> Fe <sub>x</sub> Se <sub>2</sub> (Ln = lanthanide elements)
BaZnBi <sub>2</sub>	K <sub>2</sub> NbO <sub>3-x</sub> F <sub>1-x</sub>	(Ca,La)RuAs <sub>2</sub>	La <sub>3</sub> Ru <sub>2</sub> B <sub>2</sub> N <sub>3</sub> [38]	LaRuBN [38]

Table 2. (Continued.)

LaRhBN [38]	(La, Sr)Pd <sub>2</sub> As <sub>2</sub>	LaIrBN [38]	(Sr,K)Pt <sub>2</sub> As <sub>2</sub>	(Ca,La)PtAs <sub>2</sub>
[Sr <sub>2</sub> MO <sub>3</sub> ][Ti <sub>2</sub> As <sub>2</sub> O] <sub>2</sub> [Sr <sub>2</sub> M <sub>2</sub> O <sub>3</sub> ][Ti <sub>2</sub> As <sub>2</sub> O] (M = Cr, Mn, Fe, Co, Ni, Cu)				
Square net of traditional elements		TiSiAs	SrGa <sub>2</sub>	BaGa <sub>2</sub>
NbGeAs	TaGeAs	BaNiSn <sub>3</sub>	LaMn <sub>2</sub> Sb <sub>2</sub>	La <sub>2</sub> O <sub>2</sub> Sb
Ce <sub>2</sub> O <sub>2</sub> Sb	CeAgSb <sub>2</sub>	Nd(Fe <sub>1-x</sub> Co <sub>x</sub> ) <sub>2</sub> Sb <sub>2</sub>	La <sub>2</sub> O <sub>2</sub> Te	Y <sub>2</sub> O <sub>2</sub> Bi [81]
La <sub>2</sub> O <sub>2</sub> Bi [81]	LaAgBi <sub>2</sub>	Ce <sub>2</sub> O <sub>2</sub> Bi [81]	CeMn <sub>2</sub> Bi <sub>2</sub>	CeZn <sub>2</sub> Bi <sub>2</sub>
CeCu <sub>2</sub> Bi <sub>2</sub>	CeAgBi <sub>2</sub>	Pr <sub>2</sub> O <sub>2</sub> Bi [81]	Nd <sub>2</sub> O <sub>2</sub> Bi [81]	Sm <sub>2</sub> O <sub>2</sub> Bi [81]
Eu <sub>2</sub> O <sub>2</sub> Bi	Gd <sub>2</sub> O <sub>2</sub> Bi	Er <sub>2</sub> O <sub>2</sub> Bi [81]	Yb <sub>2</sub> O <sub>2</sub> Bi	Y <sub>2</sub> Si <sub>3</sub>
La <sub>2</sub> Si <sub>3</sub>				
Other layered structure				
Honeycomb lattice		Mg <sub>2</sub> PtSi	SrPdAs	(Sr,Y)PtAs
SrPtSb	BaPtSb	SrPdSb	YPtAs	LaPtSb
MgAgAs	SrCoP	SrNiP	SrCuP	SrPdP
SrAgP	SrIrP	SrAuSn	SrPtSn	SrAuSb
CaAuP	SrAuP	SrAuSi		
Mn <sub>2</sub> Si <sub>2</sub> structure		V <sub>2</sub> P <sub>3</sub>	V <sub>2</sub> P <sub>2</sub> N <sub>2</sub>	V <sub>2</sub> As <sub>2</sub> N
Nb <sub>2</sub> Si <sub>3</sub>	Ta <sub>2</sub> Si <sub>3</sub>	V <sub>2</sub> Ge <sub>3</sub>	Nb <sub>2</sub> Ge <sub>2</sub> N <sub>2</sub>	Ta <sub>2</sub> Ge <sub>2</sub> N <sub>2</sub>
Nb <sub>2</sub> Ge <sub>2</sub> C <sub>2</sub>	Zr <sub>2</sub> Pb <sub>3</sub>	Ta <sub>2</sub> Ir <sub>2</sub> O		
Mg <sub>2</sub> AlSi <sub>3</sub> structure		LiMg <sub>2</sub> Si <sub>3</sub>	NaMg <sub>2</sub> Si <sub>3</sub>	SrMg <sub>2</sub> Si <sub>3</sub>
BaMg <sub>2</sub> Si <sub>3</sub>	YMg <sub>2</sub> Si <sub>3</sub>	BaGe <sub>2</sub> Mg <sub>2</sub>		
B, Si layer		CrB <sub>2</sub>	MnB <sub>2</sub>	FeB <sub>2</sub>
ZnB <sub>2</sub>	SrNiSi <sub>3</sub>	Ba(Si <sub>1-x</sub> Zn <sub>x</sub> ) <sub>2</sub>	Ba(Si <sub>1-x</sub> Ni <sub>x</sub> ) <sub>2</sub>	SiTe <sub>2</sub>
SmGa <sub>2-x</sub> Si <sub>x</sub>	CaCuSi	YbGa <sub>0.9</sub> Si <sub>1.1</sub>	LaGa <sub>2-x</sub> Si <sub>x</sub>	La <sub>2</sub> AlSi <sub>2</sub>
CaSrSi <sub>4</sub>	Ba <sub>2</sub> Ga <sub>0.7</sub> Si <sub>1.3</sub>	Sr <sub>1-x</sub> Na <sub>x</sub> Al <sub>2</sub> Si <sub>2</sub>	Ba <sub>1-x</sub> K <sub>x</sub> Al <sub>2</sub> Si <sub>2</sub>	Ca <sub>1-x</sub> Na <sub>x</sub> Al <sub>2</sub> Si <sub>2</sub> [45]
SrAl <sub>2-x</sub> Si <sub>2+x</sub>	BaSi <sub>2</sub> (HP phase)			
Skutterdite structure		Sr <sub>2</sub> RhGe <sub>1.5</sub> Se <sub>1.5</sub>	Ba <sub>2</sub> IrGe <sub>1.5</sub> Si <sub>1.5</sub>	Ba <sub>2</sub> IrGe <sub>1.5</sub> Se <sub>1.5</sub>
BaRh <sub>2</sub> Ge <sub>4</sub> Se <sub>6</sub> [82]	BaRh <sub>2</sub> Ge <sub>4</sub> Se <sub>6</sub> [82]	BaIr <sub>2</sub> Ge <sub>4</sub> Se <sub>6</sub> [82]	BaIr <sub>2</sub> Ge <sub>4</sub> Se <sub>6</sub> [82]	La <sub>2</sub> IrGe <sub>1.5</sub> Si <sub>1.5</sub>
La <sub>2</sub> RhGe <sub>1.5</sub> Si <sub>1.5</sub>	LaFe <sub>2</sub> Bi <sub>12</sub>	LaCo <sub>2</sub> Bi <sub>12</sub>	LaNi <sub>2</sub> Bi <sub>12</sub>	LaRu <sub>2</sub> Bi <sub>12</sub>
LaRh <sub>2</sub> Bi <sub>12</sub>	LaPd <sub>2</sub> Bi <sub>12</sub>	LaIr <sub>2</sub> Bi <sub>12</sub>	LaPt <sub>2</sub> Bi <sub>12</sub>	
Others		Y(Co,Fe) <sub>2</sub> B <sub>2</sub>	Y(Co,Ni) <sub>2</sub> B <sub>2</sub>	Ca <sub>2</sub> N [83]
Ca <sub>2</sub> BN (film) [84]	Ba(Ge <sub>1-x</sub> Cu <sub>x</sub> ) <sub>2</sub>	Cr <sub>2</sub> N	CuNCN	NbCrN
TiNF	Ti <sub>2</sub> SbP	Zr <sub>2</sub> SbP	Cr <sub>2</sub> Ti	Cr <sub>2</sub> Zr
Cr <sub>2</sub> Hf	CrTaN	NiTe <sub>2</sub>	Ni(Te <sub>1-x</sub> Se <sub>x</sub> ) <sub>2</sub>	NaNiF <sub>3</sub> [85]
SmNiAs	ZrAs	Ni <sub>2</sub> Bi <sub>3</sub>	CaRuO <sub>3</sub> [86]	CaRhO <sub>3</sub> [87]
CaOsO <sub>4</sub>	SrOsO <sub>4</sub>	BaOsO <sub>3</sub>	CaRhO <sub>3</sub> (HP phase) [88]	MAN phase (M=Ti, V, Zr, Nb, Hf, Ta) (A=Si, P, S, Ga, In, Sn, Pb)
La <sub>3</sub> Ni <sub>2</sub> O <sub>7</sub>	Sr <sub>2</sub> SnO <sub>4-x</sub>	Ba <sub>2</sub> SnO <sub>4-x</sub>	SnO: F	Cu(Al,Mg)O <sub>2</sub>
SrCr <sub>2</sub> O <sub>4</sub>	LaSrMnO <sub>3</sub> H	(CuCl)LaNb <sub>2</sub> O <sub>7-x</sub> F <sub>x</sub>	Bi <sub>2</sub> WO <sub>6</sub>	Ho <sub>0.25</sub> Sr <sub>0.75</sub> FeO <sub>2-x/d</sub>
Na <sub>2</sub> La <sub>2</sub> Ti <sub>2</sub> O <sub>10-x</sub>	Sr <sub>2</sub> TiO <sub>4</sub> F <sub>2</sub>	CaCu <sub>2</sub> Cr <sub>2</sub> O <sub>12</sub>	SrCuO <sub>2</sub> : F	
Intercalation to layered structure				
FeOCl	VOCl	TiOCl	CrOCl	TiNCl ← mono-amine [55]
Zr <sub>2</sub> N <sub>2</sub> S [89]	Hf <sub>2</sub> N <sub>2</sub> S [89]	Ca <sub>2</sub> Hf <sub>2</sub> N <sub>2</sub> S	α-HfNB	Ti <sub>2</sub> PTe <sub>2</sub> ← Ag, Cu, Zn [90]
Bi <sub>2</sub> PdO <sub>4</sub> F <sub>2</sub>	Bi <sub>1.67</sub> Pb <sub>0.33</sub> PtO <sub>4</sub> F <sub>2</sub>	Pb <sub>2</sub> O <sub>4</sub> ← F	Zr <sub>2</sub> (N <sub>2</sub> O)F <sub>6-x</sub>	Ti <sub>2</sub> N <sub>2</sub> S
Ti <sub>2</sub> N <sub>2</sub> O	Hf <sub>2</sub> (N <sub>2</sub> O) <sub>2</sub> F <sub>6-x</sub>	Zr <sub>2</sub> N <sub>2</sub> O	Graphite ← KH	graphite ← B
CsLaTe <sub>2</sub> O <sub>7</sub> Cl				
perovskite and its related structure				
antiperovskite structure		Mn <sub>2</sub> AlC	Mn <sub>2</sub> SnC	Mn <sub>2</sub> AlN
Mn <sub>2</sub> CrN	Mn <sub>2</sub> ZnN [91]	AlNiC	Mn <sub>2</sub> ZnN <sub>1-x</sub> C <sub>x</sub>	Mn <sub>2</sub> InN <sub>1-x</sub> C <sub>x</sub>
Mn <sub>2</sub> Sb <sub>1-x</sub> Sn <sub>x</sub> N	Mn <sub>2</sub> Ag <sub>1-x</sub> Sn <sub>x</sub> N	Ca <sub>3</sub> SiO	Ca <sub>3</sub> SnO	Ca <sub>3</sub> PbO
SrAg <sub>3</sub> P	SrIr <sub>3</sub> P	SrAu <sub>3</sub> P	Mg <sub>2</sub> CNi <sub>4</sub>	Ca <sub>4</sub> OSb <sub>2</sub>
RhNCr <sub>3</sub>				
double perovskite structure		Ca <sub>2</sub> OsO <sub>6</sub>	Ca <sub>2</sub> MnOsO <sub>6</sub>	Ca <sub>2</sub> CoOsO <sub>6</sub>
Ca <sub>2</sub> ZnOsO <sub>6</sub>	Sr <sub>2</sub> OsO <sub>6</sub>	Sr <sub>2</sub> LiOsO <sub>6</sub> [92]	Sr <sub>2</sub> MnOsO <sub>6</sub>	Sr <sub>2</sub> FeOsO <sub>6</sub> [93]
Sr <sub>2</sub> CoOsO <sub>6</sub> [93]	Sr <sub>2</sub> NiOsO <sub>6</sub>	Sr <sub>2</sub> YO <sub>8</sub> O <sub>6</sub>	Sr <sub>2</sub> PbOsO <sub>6</sub>	Sr <sub>2</sub> DyOsO <sub>6</sub>
Ba <sub>2</sub> (Sr <sub>2-x</sub> Ca <sub>x</sub> )FeOsO <sub>6</sub>	Ba <sub>2</sub> YScOsO <sub>6</sub>	Sr <sub>2</sub> FeMoO <sub>6</sub>		
normal perovskite		SrGeO <sub>3</sub> [94]	(Ca <sub>1-x</sub> Y <sub>x</sub> )GeO <sub>3</sub>	(Sr <sub>1-x</sub> La <sub>x</sub> )GeO <sub>3</sub> [94]
CaTiO <sub>3-x</sub> H <sub>x</sub>	SrTi(O,H) <sub>3</sub> (thin film)	EuTiO <sub>3-x</sub> H <sub>x</sub>	CaVO <sub>3</sub> :H	SrVO <sub>3</sub> :H

Table 2. (Continued.)

	[95]			
SrCrO <sub>2</sub> H [96]	Sr <sub>1-x</sub> Ba <sub>x</sub> CrO <sub>2</sub> H (x = 0.1-0.4)	LnCrO <sub>3</sub> (Ln = La, Sm, Nd)	BaTiO <sub>3-x</sub>	BaZrO <sub>2</sub> :H
SrSnO <sub>3-x</sub>	SrNbO <sub>3</sub> N	BaSnO <sub>3-x</sub>	NaIrO <sub>3</sub>	Na <sub>1-x</sub> Ca <sub>x</sub> IrO <sub>3</sub>
SrMoO <sub>2.5</sub> N <sub>0.5</sub>	BaBiS <sub>3</sub>			
other perovskite		A <sub>2</sub> Ru <sub>2</sub> O <sub>5</sub>	LaBaMn <sub>2</sub> O <sub>6</sub>	YBaMn <sub>2</sub> O <sub>6</sub>
A <sub>2</sub> Cr <sub>2</sub> O <sub>5</sub>				
cuprate superconductor and its related structure				
(Sr,Eu)CuO <sub>2</sub>	(Sr,Pb)CuO <sub>2</sub>	Bi <sub>2</sub> CuO <sub>4</sub> -CaH <sub>4</sub>	Bi <sub>2</sub> CuO <sub>4</sub> -CuF <sub>2</sub>	BiOCuS
YBCO:H	(Cu <sub>2</sub> S <sub>2</sub> )Sr <sub>2</sub> Fe <sub>2</sub> O <sub>5</sub>	Sr <sub>2</sub> FeNiCu <sub>2</sub> Se <sub>2</sub> O <sub>5</sub>	Sr <sub>2</sub> Ni <sub>2</sub> Cu <sub>2</sub> Se <sub>2</sub> O <sub>5</sub>	Sr <sub>2</sub> Cu <sub>4</sub> Se <sub>2</sub> O <sub>5</sub>
Bi <sub>2</sub> YCu <sub>2</sub> O <sub>4</sub> Se <sub>2</sub>	Bi <sub>2</sub> SrCu <sub>2</sub> O <sub>4</sub> Se <sub>2</sub>	SrLa <sub>2</sub> Cu <sub>4</sub> Se <sub>2</sub> O <sub>5</sub>	A <sub>2</sub> CuO <sub>4</sub> (OH) <sub>2</sub>	LaSrTiO <sub>4</sub>
Sr <sub>2</sub> CrO <sub>4</sub>	Ca <sub>2</sub> CrO <sub>4</sub>	Sr <sub>2</sub> Cr <sub>2</sub> O <sub>7</sub>	Ca <sub>2</sub> Cr <sub>2</sub> O <sub>7</sub>	A <sub>2</sub> Cr <sub>2</sub> O <sub>10</sub> (A = Sr, Ca)
Sr <sub>2-x</sub> Na <sub>x</sub> CrO <sub>4</sub>	YCaCrO <sub>4</sub>	Sm <sub>1-x</sub> K <sub>x</sub> Mn <sub>2</sub> O <sub>5(n-1)</sub> (n = 1, ∞)		
clathrate structure				
Ca <sub>8</sub> Ni <sub>12</sub> B <sub>6</sub>	Sr <sub>2</sub> Ni <sub>12</sub> B <sub>6</sub>	KC <sub>34</sub>	Li <sub>4</sub> C <sub>60</sub>	K <sub>4</sub> C <sub>60</sub>
Ca <sub>4</sub> C <sub>60</sub>	Ba <sub>4</sub> C <sub>60</sub>	Mg <sub>2</sub> C <sub>60</sub>	Ba <sub>3</sub> C <sub>60</sub> [97]	Ca <sub>3</sub> AlN
Li <sub>4</sub> Si <sub>136</sub>	Na <sub>4-x</sub> Si <sub>136</sub> [98]	K <sub>4</sub> Si <sub>136</sub>	Mg <sub>2</sub> Si <sub>136</sub>	Ca <sub>3</sub> Si <sub>136</sub>
Ba <sub>4</sub> Si <sub>136</sub>				
Pyrite structure				
FeS <sub>2-x</sub>	PtSb <sub>2</sub>	(Pt,Pd)Sb <sub>2</sub>	(Pt,Ir)Sb <sub>2</sub> [99]	Pt(Sb,As) <sub>2</sub>
Pt(Sb,Te) <sub>2</sub>	Pt(Sb,Sn) <sub>2</sub>	(Pt,Au)Sb <sub>2</sub>	IrSb <sub>2</sub>	
other materials				
carbide		CaC <sub>2</sub>	BaC <sub>2</sub>	YC <sub>2</sub>
nitride		Ti <sub>3</sub> N <sub>4</sub>	Zr <sub>3</sub> N <sub>4</sub>	Hf <sub>3</sub> N <sub>4</sub>
VCrN	Mg <sub>3</sub> AlN	MnN (thin film)	CoN (thin film)	
oxide		TiO <sub>2</sub> (anatase) [100]	Cr <sub>2</sub> O	Cr <sub>2</sub> O <sub>2n-1</sub> (n = 3, 4)
MnO <sub>2</sub> (nanosheet)	Co <sub>3</sub> O <sub>4</sub> : H	NiO: H	GeO <sub>2</sub> (HP phase)	h-WO <sub>3</sub> : H
BaScO <sub>2</sub> H	Ti <sub>3</sub> O <sub>4.8</sub> N <sub>0.4</sub>	Ti <sub>4</sub> O <sub>6.96</sub> N <sub>0.01</sub>	(Ir,Ru)O <sub>2</sub>	CuAl(O,N) <sub>2</sub>
LaSiO <sub>3</sub> H	NaTi <sub>2</sub> O <sub>4</sub>	Bi <sub>4</sub> Ti <sub>3</sub> O <sub>12</sub> : H	Sc <sub>2</sub> Ti <sub>3</sub> O <sub>12</sub>	Cs <sub>0.7</sub> Ti <sub>1.425</sub> O <sub>4</sub> : F
La <sub>2</sub> Ti <sub>2</sub> O <sub>7-x</sub>	Ba-Ti-O glass + e	Y <sub>2</sub> Ti <sub>2</sub> O <sub>7-x</sub>	Sr <sub>2</sub> Ti <sub>2</sub> F <sub>2</sub>	Cr <sub>1-x</sub> Ti <sub>x</sub> O <sub>2</sub>
Cr <sub>1-x</sub> V <sub>x</sub> O <sub>2</sub>	nSrO · V <sub>2</sub> O <sub>5</sub> (n = 4, 2, 1)	Bi <sub>2</sub> V <sub>8</sub> O <sub>16</sub>	CaV <sub>2</sub> O <sub>5</sub>	LiV <sub>2</sub> O <sub>5</sub>
LaVO <sub>3</sub> : H	LaVO <sub>4</sub> : H	NaCr <sub>7</sub> O <sub>4</sub> [101]	Ca <sub>1-x</sub> Na <sub>x</sub> Cr <sub>2</sub> O <sub>4</sub>	CaCr <sub>2</sub> O <sub>4</sub>
BaCr <sub>2</sub> O <sub>4</sub>	SrCu <sub>3</sub> Cr <sub>2</sub> O <sub>12</sub>	BaCu <sub>3</sub> Cr <sub>2</sub> O <sub>12</sub>	YCu <sub>3</sub> Cr <sub>2</sub> O <sub>12</sub>	LaCu <sub>3</sub> Cr <sub>2</sub> O <sub>12</sub>
CeCu <sub>3</sub> Cr <sub>2</sub> O <sub>12</sub>	LuCu <sub>3</sub> Cr <sub>2</sub> O <sub>12</sub>	EuCu <sub>3</sub> Cr <sub>2</sub> O <sub>12</sub>	BiCu <sub>3</sub> Cr <sub>2</sub> O <sub>12</sub>	Cr <sub>1-x</sub> Mn <sub>x</sub> O <sub>2</sub>
Ag <sub>2</sub> MnO <sub>4</sub>	Pb <sub>2</sub> MnO <sub>4</sub> : F	LiMn <sub>2</sub> O <sub>4</sub>	Sr <sub>3</sub> Co <sub>2</sub> O <sub>6</sub> [102]	Ba <sub>3</sub> Co <sub>2</sub> O <sub>6</sub> (CO <sub>3</sub> ) <sub>2</sub>
Sr <sub>1-x</sub> Y <sub>x</sub> Co <sub>2</sub> O <sub>6</sub> [103]	Sr <sub>1-x</sub> Ca <sub>x</sub> Co <sub>2</sub> O <sub>6</sub>	Sr <sub>3</sub> Co <sub>2</sub> Zn <sub>2</sub> O <sub>6</sub>	Sr <sub>3</sub> Co <sub>2</sub> Fe <sub>2</sub> O <sub>6</sub>	Y-ZrO <sub>2-x</sub>
Sr <sub>2</sub> Nb <sub>2</sub> O <sub>4</sub> : F	Ba <sub>3</sub> Nb <sub>2</sub> O <sub>13-x</sub>	Ba <sub>2</sub> Nb <sub>2</sub> O <sub>4</sub> : F	PbMoO <sub>4</sub> : N	Ca <sub>2</sub> Nb <sub>2</sub> O <sub>6</sub> : F <sub>1-x</sub>
Sr <sub>2</sub> NbO <sub>4</sub> N	Li <sub>2</sub> Ir <sub>1-x</sub> Ru <sub>x</sub> O <sub>3</sub> [104]	La <sub>4</sub> Ru <sub>2</sub> O <sub>10</sub>	Bi <sub>3</sub> Ru <sub>2</sub> O <sub>11</sub>	Sr <sub>2</sub> Ru <sub>2</sub> O <sub>7</sub> : C <sub>12</sub>
Sr <sub>2</sub> Ru <sub>2</sub> O <sub>7</sub> F <sub>2</sub>	LaCu <sub>3</sub> Ru <sub>2</sub> O <sub>12</sub>	CdRh <sub>2</sub> O <sub>4</sub> [105]	Cd <sub>1-x</sub> Na <sub>x</sub> Rh <sub>2</sub> O <sub>4</sub>	La <sub>2</sub> Pd <sub>2</sub> O <sub>7</sub>
BaSbO <sub>3</sub>	CsW <sub>2</sub> O <sub>5</sub>	SrWO <sub>2</sub> N	LiOsO <sub>3</sub> [106]	Na <sub>1-x</sub> Ca <sub>x</sub> OsO <sub>3</sub>
KOsO <sub>3</sub>	Ca <sub>2</sub> LiOsO <sub>5</sub> [107]	Ca <sub>2</sub> Os <sub>2</sub> O <sub>7</sub>	Ca <sub>3</sub> Os <sub>2</sub> O <sub>12</sub>	Ba <sub>2</sub> OsO <sub>5</sub>
Bi <sub>2</sub> OsO <sub>6</sub>	Pb <sub>2</sub> Os <sub>2</sub> O <sub>7</sub>	Pb <sub>2</sub> Ir <sub>2</sub> O <sub>7</sub>	Ag <sub>2</sub> BiO <sub>3</sub>	Ba <sub>1-x</sub> K <sub>x</sub> BiO <sub>3</sub> (junction)
chalcogenide				
SrPt <sub>2</sub> S	BaPt <sub>2</sub> S	Cu <sub>2</sub> S	Cu(S,As)	KMo <sub>3</sub> S <sub>3</sub>
Cu <sub>2</sub> Bi <sub>2</sub> Se <sub>3</sub>	Cu <sub>2</sub> Bi <sub>2</sub> Te <sub>3</sub>	InV <sub>2</sub> Se <sub>3</sub>	Pt <sub>2</sub> Bi <sub>2</sub> Se <sub>3</sub>	KMo <sub>3</sub> Se <sub>3</sub>
pnictide		CoP	NiP	(Co,Ni)P
La <sub>3</sub> Ir <sub>3</sub> P <sub>12</sub>	Mg <sub>3</sub> AlP	BaNi <sub>2</sub> P <sub>4</sub>	BaPd <sub>2</sub> P <sub>4</sub>	(Y <sub>1-x</sub> Eu <sub>x</sub> )P
SrPt <sub>2</sub> Sb	BaPt <sub>2</sub> Sb	La <sub>3</sub> TiSb <sub>5</sub>	La <sub>3</sub> VSb <sub>5</sub>	La <sub>3</sub> CrSb <sub>5</sub>
La <sub>3</sub> MnSb <sub>5</sub>	La <sub>3</sub> TiBi <sub>4</sub>	La <sub>3</sub> VBi <sub>5</sub>	La <sub>3</sub> CrBi <sub>5</sub>	La <sub>3</sub> ZtBi <sub>4</sub>
La <sub>3</sub> HfBi <sub>5</sub>				
silicide, germanide, stannide, plumbide				
Mg <sub>2</sub> AlSi <sub>6</sub>	Mg <sub>1.5</sub> Ga <sub>0.5</sub> Si	(Ca, Mg)Si <sub>2</sub>	Fe <sub>2</sub> Al <sub>2</sub> Si <sub>4</sub>	MgAl <sub>2</sub> Si <sub>2</sub>
BaPdSi	SrPdSi	Li <sub>3</sub> PtSi <sub>3</sub>	BaMg <sub>2</sub> Si <sub>2</sub>	Ti <sub>2</sub> Ru <sub>3</sub> Si <sub>4</sub>
BaMg <sub>2</sub> Pb <sub>2</sub>		BaMg <sub>2</sub> Ge <sub>2</sub>	Rh <sub>3</sub> Ge <sub>7</sub>	BaMg <sub>2</sub> Sn <sub>2</sub>
halide				
Zr <sub>3</sub> (N <sub>4</sub> H) <sub>2</sub> F <sub>6</sub>	CsSnI <sub>3</sub>	K <sub>1-x</sub> TiF <sub>4</sub>	K <sub>1-x</sub> TiF <sub>4</sub>	Zr <sub>3</sub> (N <sub>2</sub> Na)F <sub>6</sub>
metal		ZrMn <sub>2</sub>	Zr <sub>3</sub> Pb <sub>4</sub>	