Ocean acidification: a biogeological perspective

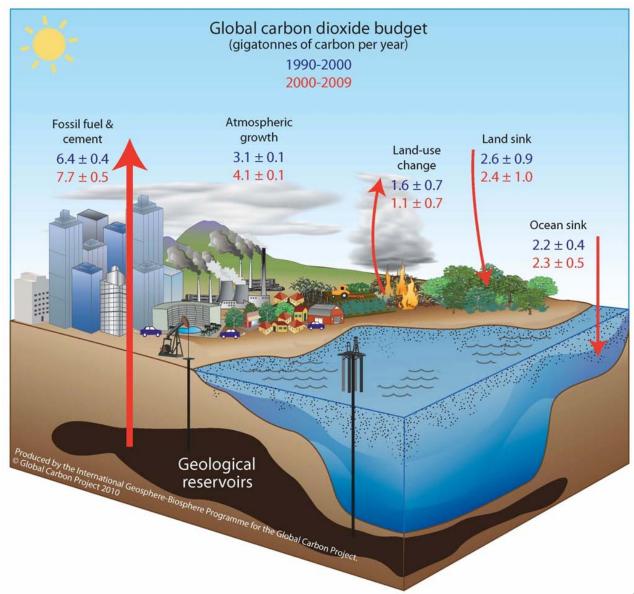
Jelle Bijma (AWI, Bremerhaven, Germany)

- Ocean acidification: present and future
- Why a biogeological perspective?
- Ocean acidification in the past
- Consequences for Biodiversity

Artist Impression of the Human Perturbation of the Carbon Cycle



Anthropogenic Global Carbon Dioxide Budget







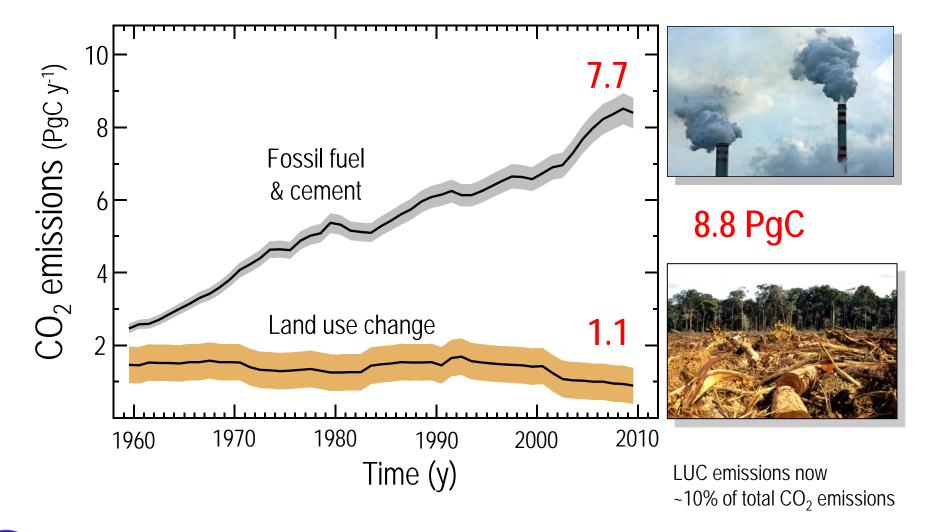




CO₂ Emissions from Land Use Change (1960-2009)

 $[1 \text{ Pg} = 1 \text{ Petagram} = 1 \text{ Billion metric tonnes} = 1 \text{ Gigatonne} = 1x10^{15}g]$

Average (2000-2009)





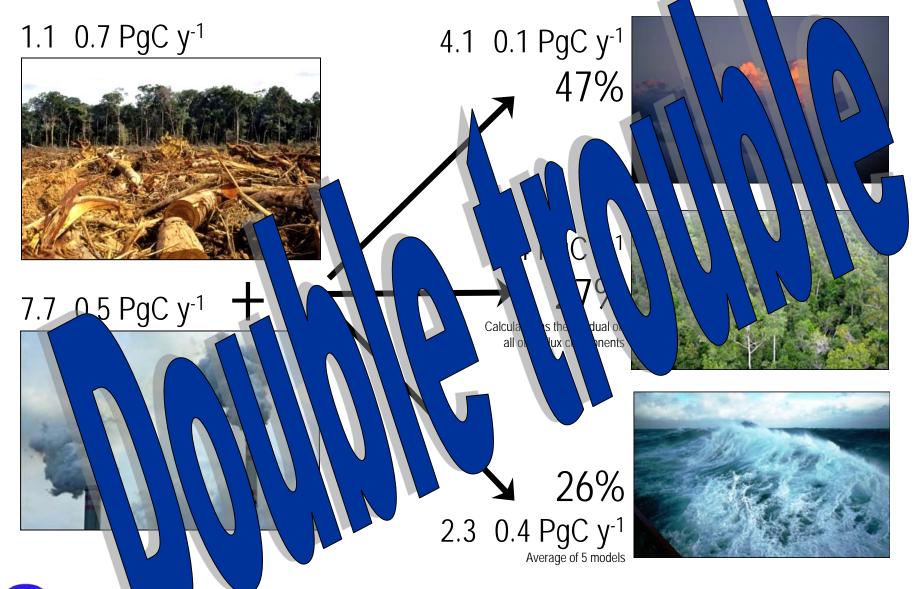








Fate of Anthropogenic CO₂ Emissions (2020-2009)







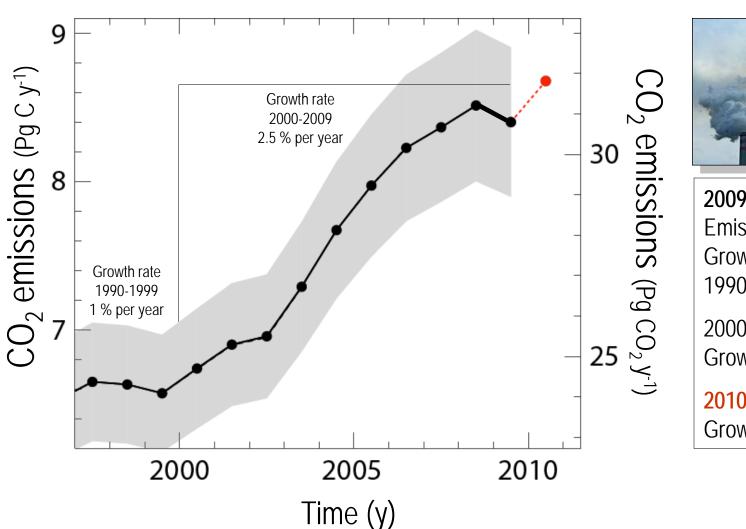






Fossil Fuel CO₂ Emissions

 $[1 \text{ Pg} = 1 \text{ Petagram} = 1 \text{ Billion metric tonnes} = 1 \text{ Gigatonne} = 1 \text{x} 10^{15} \text{g}]$





2009:

Emissions: 8.4 0.5 PgC

Growth rate: -1.3% 1990 level: +37%

2000-2008

Growth rate: +3.2%

2010 (projected):

Growth rate: >3%



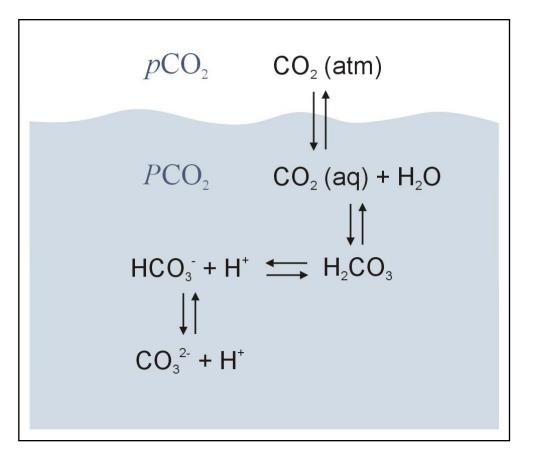








The marine carbonate system



CO₂ (aq): aqueous carbon dioxide

HCO₃: bicarbonate ion

CO₃²: carbonate ion

H₂CO₃: carbonic acid

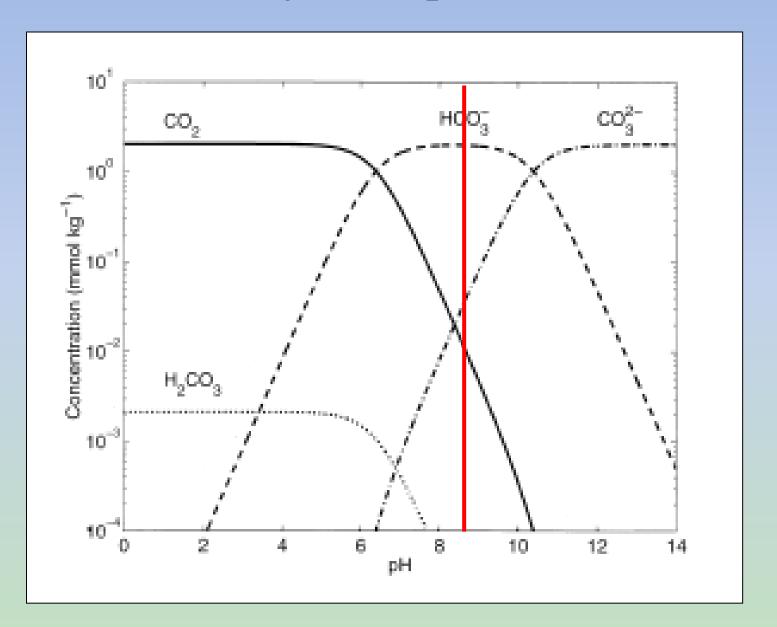
 ΣCO_2 or DIC or TCO_2 :

Total dissolved inorganic carbon

$$\Sigma CO_2 = [HCO_3^-] + [CO_3^2] + [CO_2(aq)] + [H_2CO_3]$$

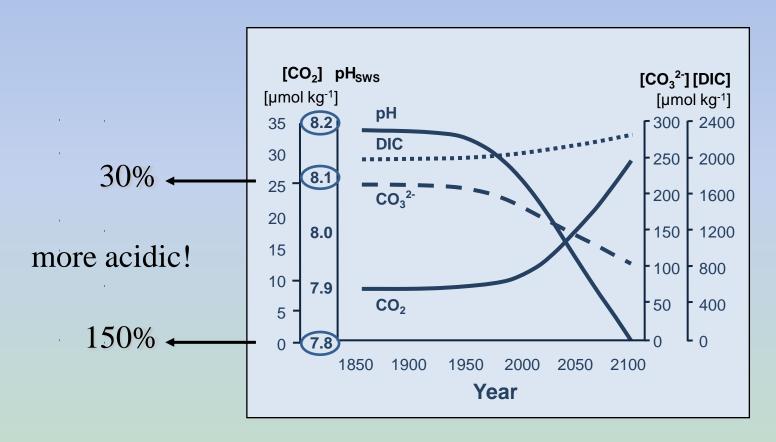
~90% ~10% <1%

"Bjerrum plot"

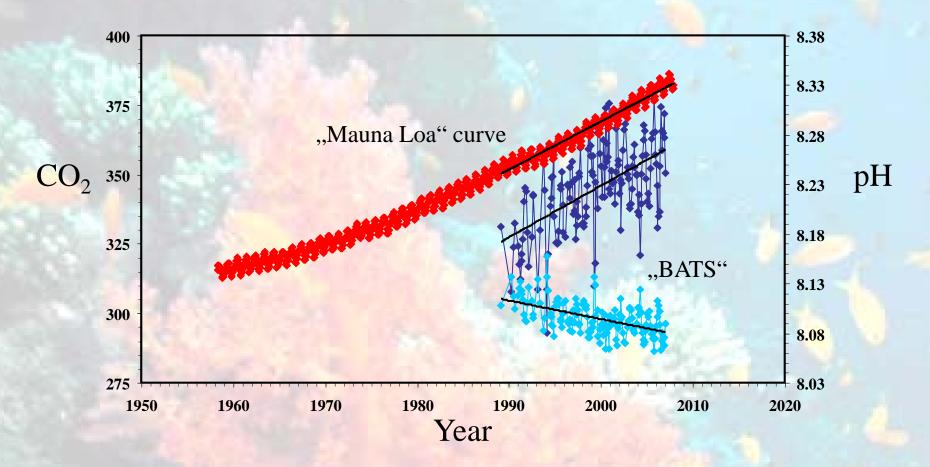


Ocean Acidification

Changes in surface ocean chemistry based on the IS92a scenario IPCC report 1995 (linear increase from 6.3 GtC yr⁻¹ to 20 GtC yr⁻¹ in the year 2100).



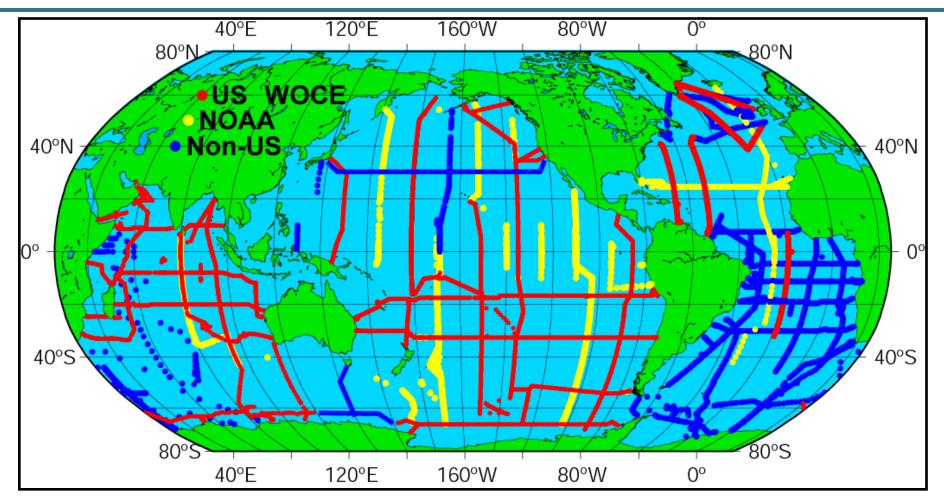
What we know about ocean CO_2 chemistry ... from time series stations



Courtesy: Richard A. Feely
NOAA/Pacific Marine Environmental Laboratory

What we know about ocean CO_2 chemistry

...from field observations



WOCE/JGOFS/OACES Global CO2 Survey

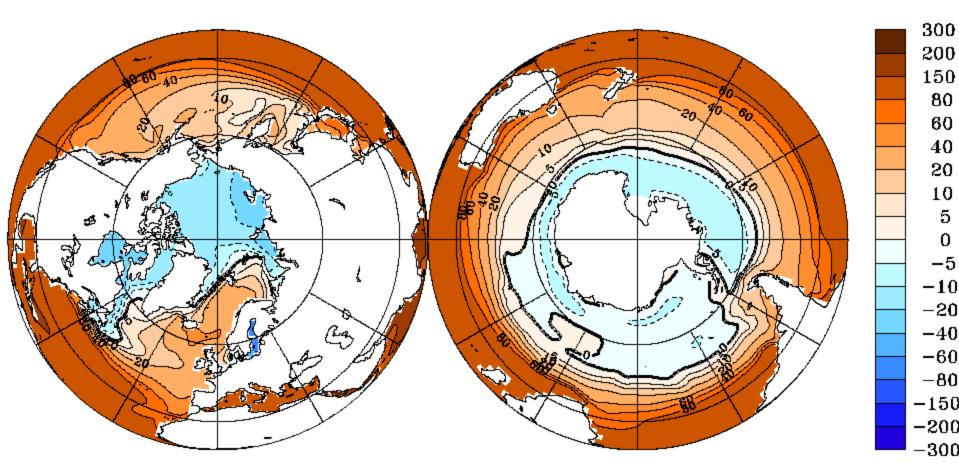
~72,000 sample locations collected in the 1990s

DIC 2 μ mol kg⁻¹ TA 4 μ mol kg⁻¹

4 μ mol kg⁻¹ Sabine et al (2004)

Undersaturation is strongest in the high latitudes

Aragonite undersaturation $\Delta[CO_3^2]_{Arag}$ at $2xCO_2$



^{*}Model approach assuming a simulation with +1% increase per year (model results only)

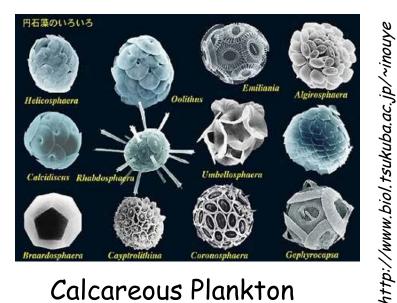
-10 -20-40-60-80

- Decrease in pH 0.1 over the last two centuries (30% increase in acidity; decrease in carbonate ion of about 16%)
- How will this impact marine organisms and ecosystems?



Corals

Photo: Missouri Botanical Gardens



Calcareous Plankton



Bivalve juvenile stages can also be sensitive to carbonate chemistry



Control $\Omega_A = 1.5$



Hard shell clam Mercenaria

 Common in soft bottom habitats

Used newly settled clams

- · Size 0.3 mm
- Massive dissolution within 24 h in undersaturated water; shell gone w/in 2 wks
- Dissolution is source of mortality in estuaries & coastal habitats

Potential impacts of high CO_2 on marine fauna

- Adverse effects on reproductive success
 - Decreased fertilization rates (sea urchins, bivalves)
 - Increased juvenile mortality (bivalves, sea urchins, copepods, fish larvae)
- Reduced growth in adults (sea urchins, bivalves)
- Impaired oxygen transport (squid)
- Reduced metabolism/scope for activity (squid)







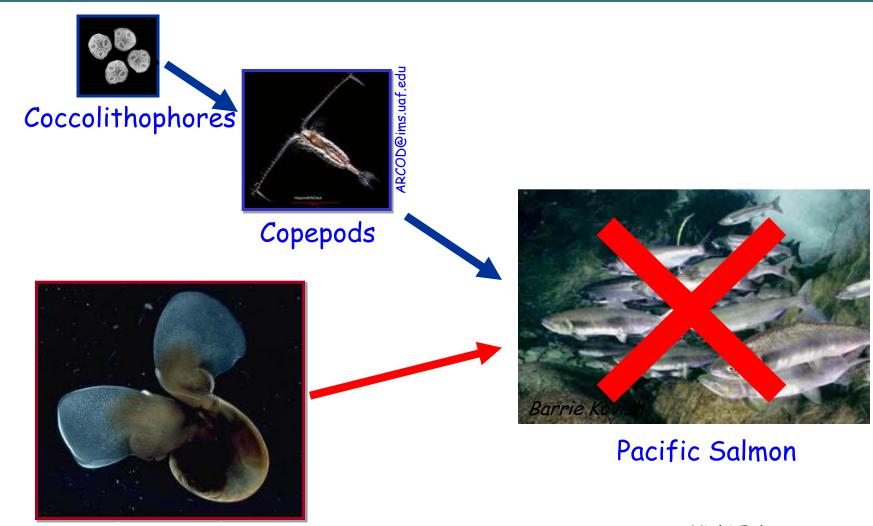
Potential Ecosystem Responses

Changes in relative abundance & distribution of calcifying species

- Non-calcifying species may outcompete calcifiers
- Geographical ranges of calcifying species may shift
- Vertical depth distributions of calcifying species may shoal with decreasing CaCO₃ saturation state

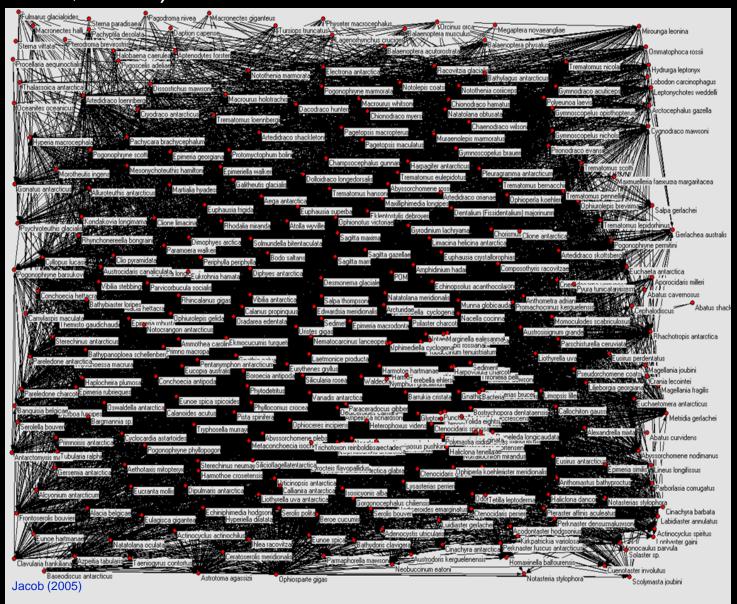
Changes in food webs and other species interactions

Potential Effects on Open Ocean Food Webs



Vicki Fabry

Weddell Sea Food Web: 489 species (incl 62 autotrophs, >16000 trophic links (Jacob, 2005)



Potential Ecosystem Responses

Changes in relative abundance & distribution of calcifying species

- > Non-calcifying species may outcompete calcifiers
- > Geographical ranges of calcifying species may shift
- ➤ Vertical depth distributions of calcifying species may shoal with decreasing CaCO₃ saturation state

Changes in food webs and other species interactions

Impacts on biogeochemical cycles

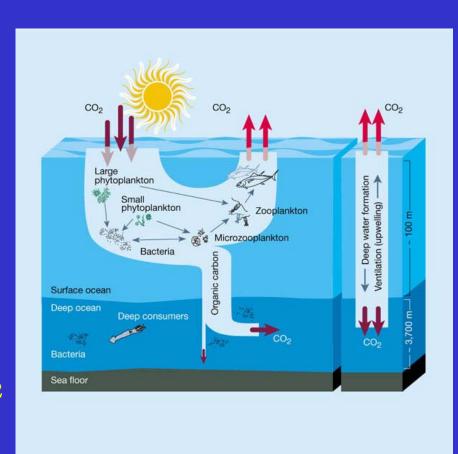
- Speciation of nutrients and trace metals
- \triangleright Changes in cycling of carbon and CaCO₃ within oceans (e.g. "ballasting")
- Changes in the "microbial loop"
- Feedbacks to climate

The global carbon cycle is largely driven by biology: How will the "biological pump" respond to OA?

What happens if biology is turned of?

The "Strangelove ocean":

- The biological pump stops
- The surface-deep CO₂ gradient disappears
- Within 250 yrs atmospheric CO₂ increases 2.4 times



see: Maier-Reimer, Mikolajewicz and Winguth (1996); Zeebe and Westbroek (2003)

Wrap up

- Oceans are stabilising global warming (but very slowly)
- At the same time are oceans acidifying (very fast)
- Society is facing double trouble....

Ocean acidification: a biogeological perspective

Jelle Bijma (AWI, Bremerhaven, Germany)

- Ocean acidification: present and future
- Why a biogeological perspective?
- Ocean acidification in the past
- Consequences for Biodiversity

Biological aspects

Real world

- comprises the actual complexity of the chemical, biological and ecological systems and interactions between them

Real time

- capture the time component inherent to carbonate perturbation and physiological and ecosystem response

Limitations

- gradual change makes it difficult to identify responses
- complexity of biology itself
- difficulty to capture longer term processes such as ecological adaptation, evolution and, biogeochemical cycles
- no information on recovery processes

Why paleo?

The farther backward you can look, the farther forward you are likely to see." Winston Churchill

- What has happened can happen (e.g. perturbation of ocean chemistry)
- Long-term (natural) context for recent changes
- Investigate the impact on biogeochemical cycles
- Reduced complexity (integration of space and time)
- Different time scales (historical/sub-recent, G-IG, deep time,....)
- Process of recovery
- Different scenarios as case and sensitivity studies and testbeds for models

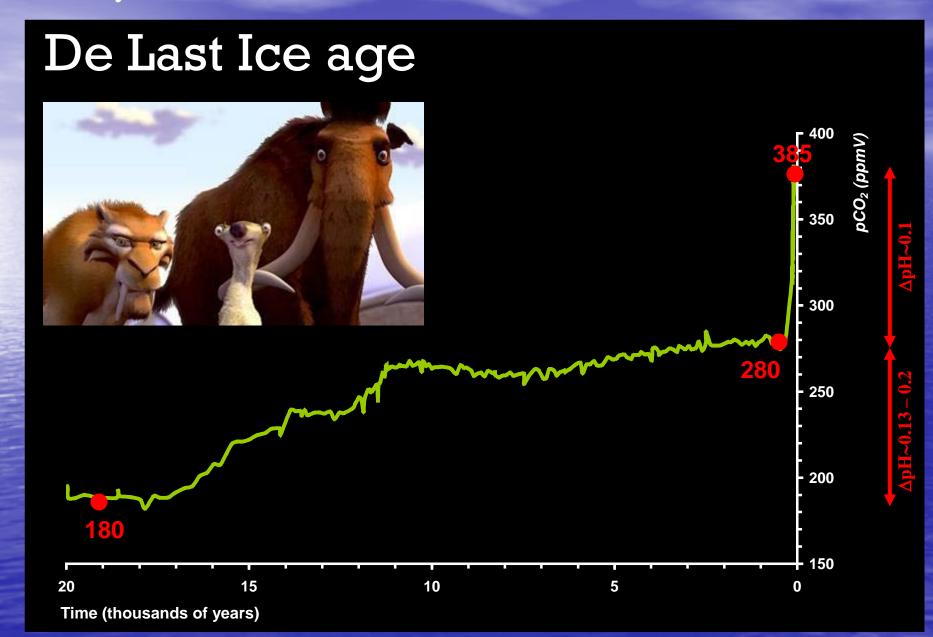
Limitations

- limited biological information (hard parts and biomarkers)
- limited by accuracy of proxy reconstructions
- restrictions on temporal and spatial resolution
- no perfect analogues

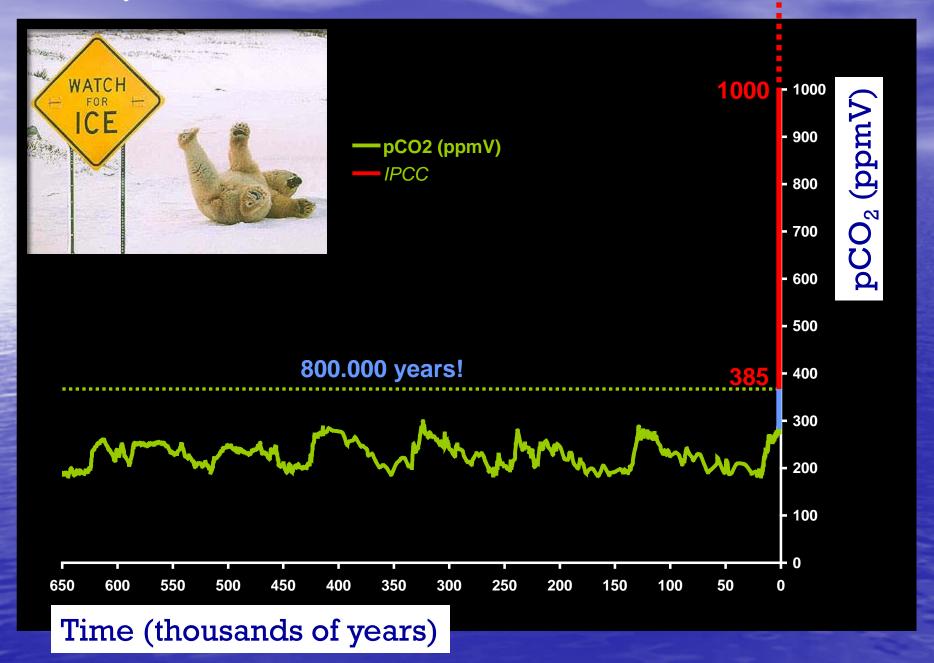
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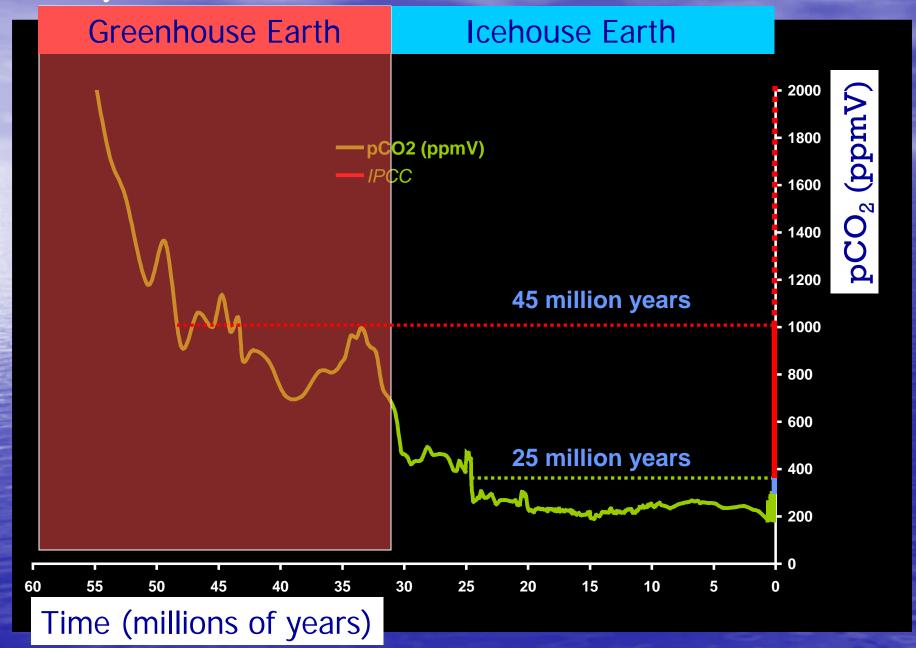
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Courtesy: Henk Brinkhuis

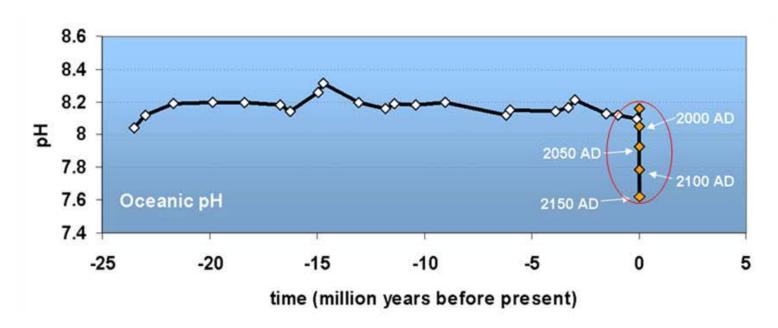


Courtesy: Henk Brinkhuis

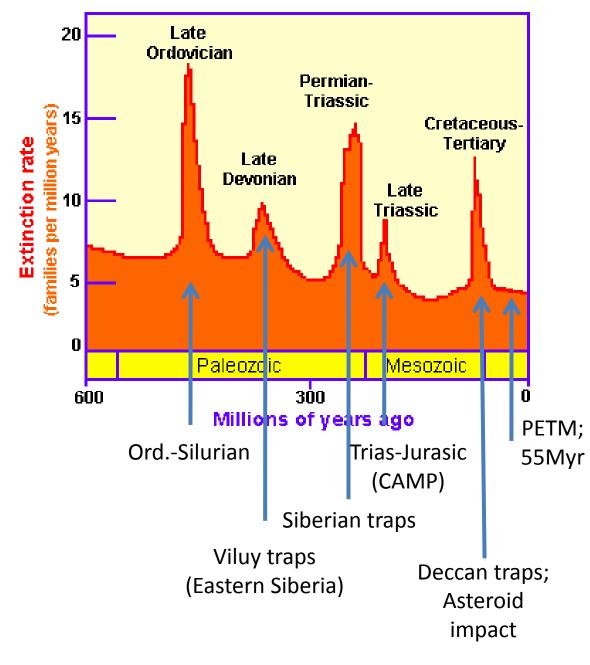


Oceans are Acidifying Fast

Changes in pH over the last 25 million years



It is happening now, at a rate and to a level not experienced by marine organisms for ~ 20MY



Carbon pertubation (symtoms):

- ➤ Global warming
- ➤Ocean acidification
- **≻**Anoxia

Evidence:

- ➤ Elevated pCO₂ (global warming)?
- ➤ Reduced pH?
- \triangleright Reduced Ω ?
- ➤ Anoxia?

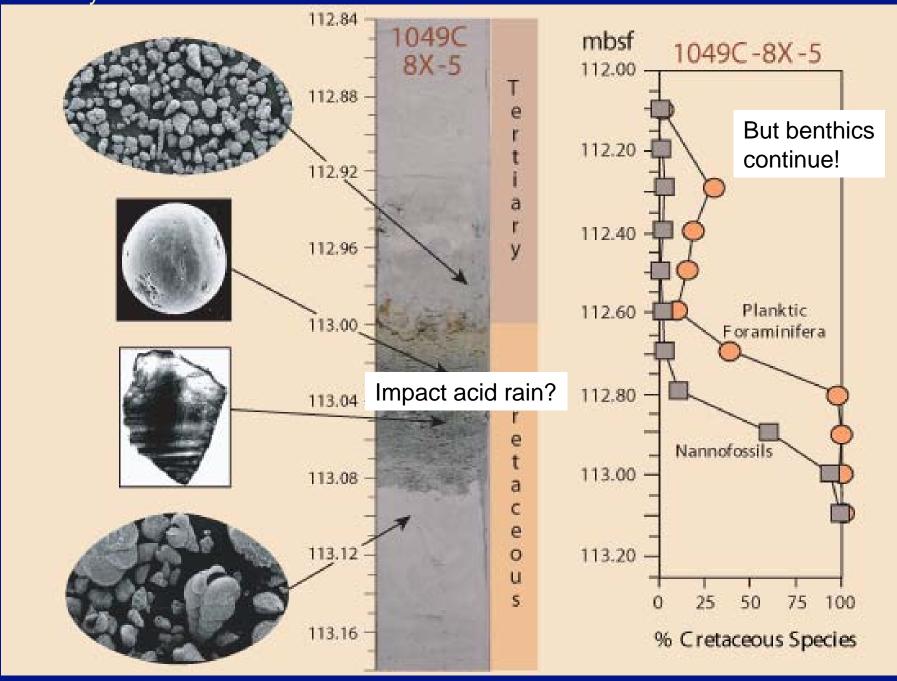


Response of marine biota to OA and climate change

• Strong perturbation at a very fast rate \rightarrow K/T impact (major planktonic extinction)

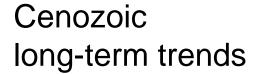
Hole 1259B 13R, 37-60 cm- boundary interval Paleogene ejecta Cretaceous 0.5 cm reflected light

Courtesy: Brian Huber



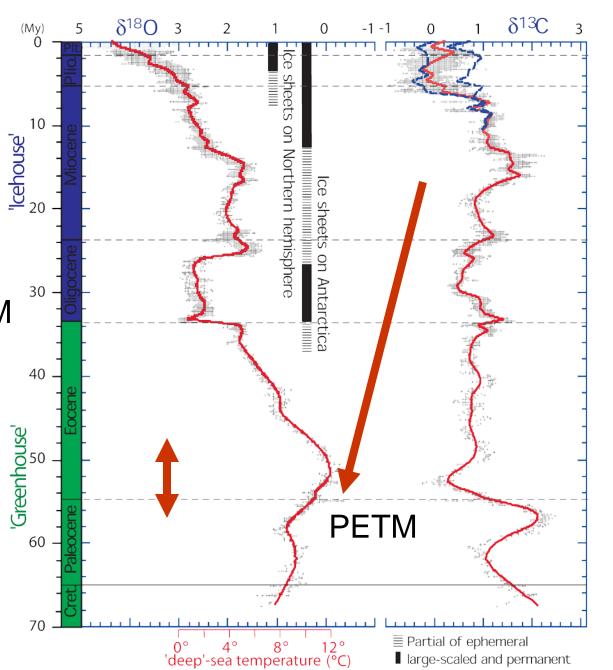
Response of marine biota to OA and climate change

- Strong perturbation at a fast rate \rightarrow K/T impact (major planktonic extinction)
- Strong perturbation at a "moderate" rate → PETM (major benthic extinction)



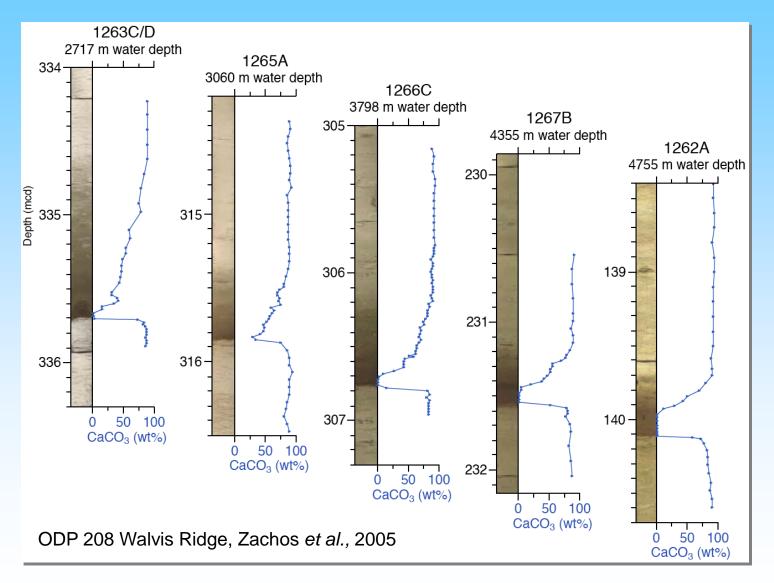
Early Eocene: warm superimposed: PETM

One of the largest benthic mass Extinction in Earth history



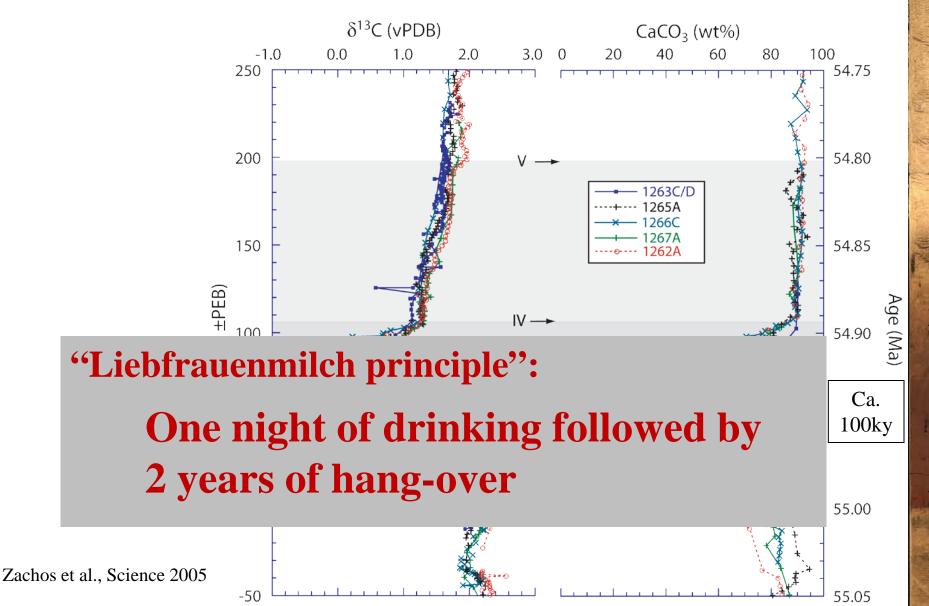
Zachos et al., 2001

Ocean Carbonate; Walvis Ridge ODP Leg 208



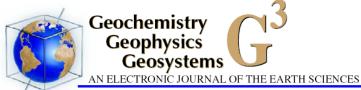
"Carbonate compensation": as lysocline is rising it destroys benthic habitats

Oceanic recovery. Walvis Ridge ODP Leg 208



Response of marine biota to OA and climate change

- Strong perturbation at a very fast rate \rightarrow K/T impact (major planktonic extinction)
- Strong perturbation at a "moderate" rate → PETM (major benthic extinction)
- Small perturbation at a slow rate → Neogene, G-IG (acclimation/adaptation)



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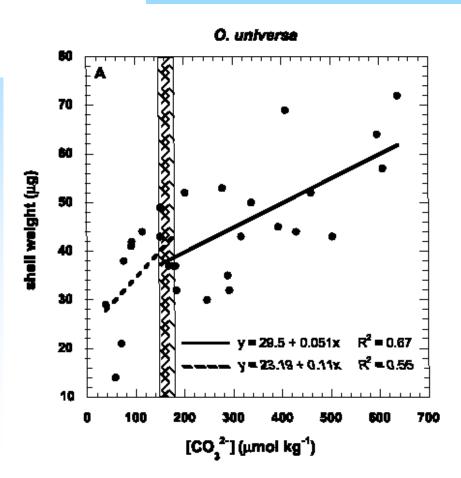
Impact of the ocean carbonate chemistry on living foraminiferal shell weight: Comment on "Carbonate ion concentration in glacial-age deep waters of the Caribbean Sea" by W. S. Broecker and E. Clark

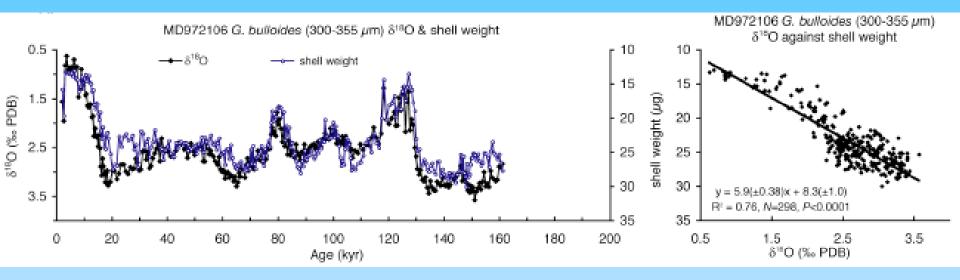
Jelle Bijma, Bärbel Hönisch, and Richard E. Zeebe

Alfred Wegener Institute, Am Handelshafen 12, Bremerhaven, D-27570, Germany (ibijma@awi-bremerhaven.de; bhoenisch@awi-bremerhaven.de; rzeebe@awi-bremerhaven.de)

 $\Delta [CO_3^{2-}]_{G-IG} \rightarrow 100 \mu mol \ kg^{-1}$

ca. 15% shell weight change



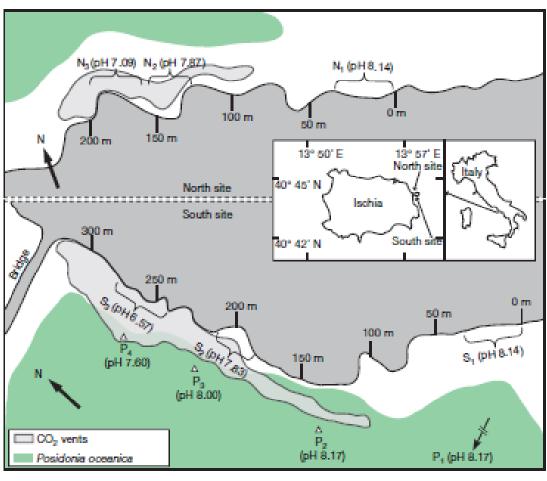


Moy, 2005

 $\Delta [\text{CO}_3^{2-}]_{\text{G-IG}} \rightarrow 100 \mu \text{mol kg}^{-1}$ ca. 50% shell weight change!

CO₂ Vents: "Windows" into High CO₂ Ocean to Assess Ecosystem Impacts





CO2 Vents: "Windows" into High CO2 Ocean to Assess Ecosystem Impacts





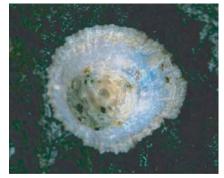
Hall-Spencer et al. Nature (2008)

Studies in the shallow waters of the Mediterranean and deep-sea show:

- total loss of some calcareous species
- reduced biodiversity
- "regime shifts": totally different ecosystems

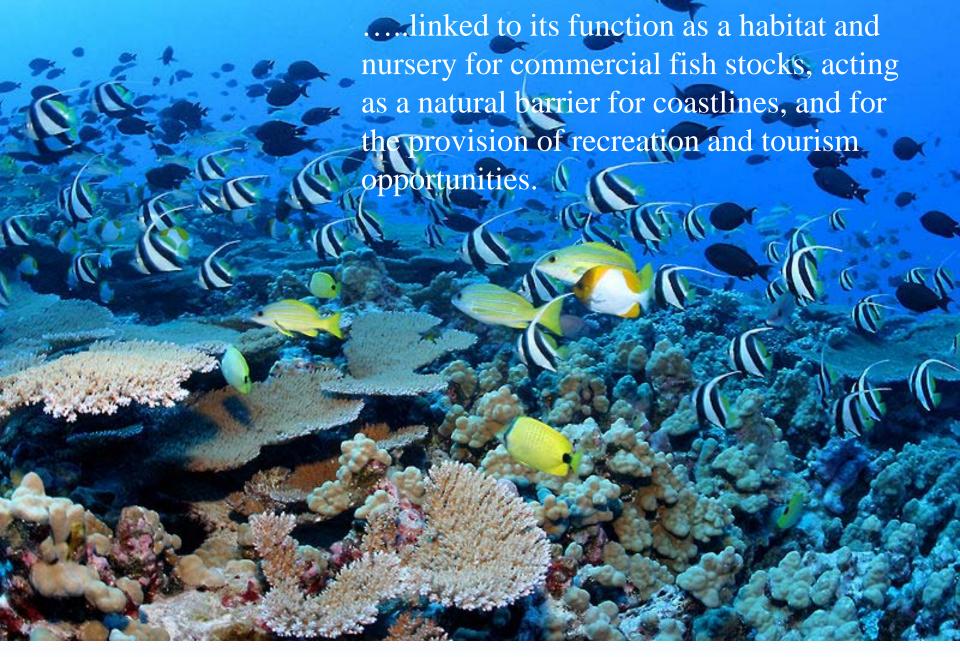
e.g. Sea grass benefit but so do invasive species





Response of marine biota to OA and climate change

- Strong perturbation at a very fast rate \rightarrow K/T impact (major planktonic extinction)
- Strong perturbation at a "moderate" rate → PETM (major benthic extinction)
- Small perturbation at a slow rate → Neogene, G-IG (acclimation/adaptation)
- Strong perturb. at a fast rate → Anthropocene: decrease in species richness → breakdown of ecosystems → extinction?



A conservative value of \$100,000 km⁻² y⁻¹ (Burke and Maidens, 2004), the global economic value associated with reefs is in the order of \$30 billion yr⁻¹.



Conclusions

- ocean acidification is ongoing and future changes are very well predictable
- organismal response poor knowledge but growing (mostly calcifiers)
- ecosystem response not known
- evolutionary capability completely unknown
- no perfect analogue to the present rate of change is unprecedented
- Earth history tells us that the combination of ocean acidification, global warming and anoxia is a deadly mix

Plymouth Marine Laboratory





Marine Matters

Is there a tipping point in ocean acidification which should be avoided?



Carol Turley
Joint EPOCA, BIOACID UKOARP Workshop on Ocean
Acidification



Potential Vulnerabilities in Relation to Human Life spans

